



New developments 1 – TrisKem

UGM - 21/09/2018

Jesus College - Cambridge (UK)

Aude Bombard

Radiopharmacy
and
Nuclear Medicine

Environment and
Bioassay

Geochemistry
and
Metals Separation

Decommissioning

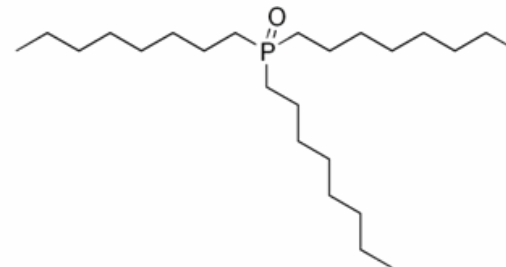


Environmental field

- **TK200 Resin**
 - Actinides separation
- **TK201 – TK202 Resins**
 - Tc/Mo separation
- **Updates**
 - Cs Resins (AMP-PAN/KNiFC-PAN)
 - TK100 resin
 - TK400 resin

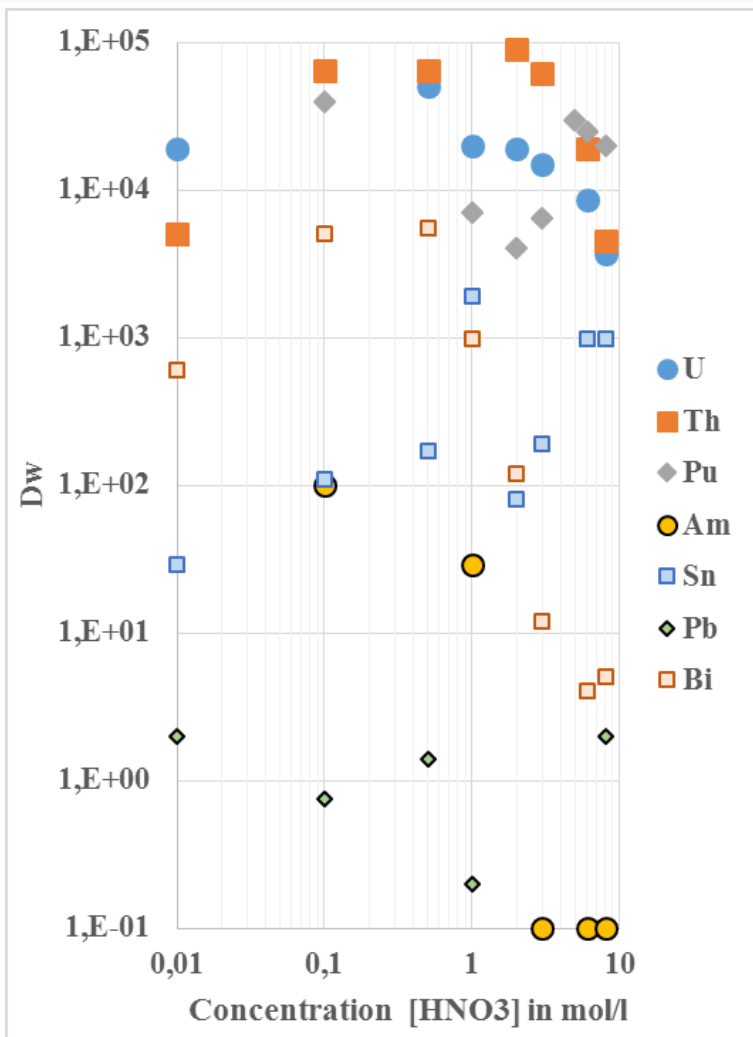


- TOPO based Resin
- Impregnated
- Well studied in Liquid-liquid systems for various metals
- Use in nuclear industry and radiotoxicology for actinides extraction[1]
- Preconcentration and sequential separation of actinides from pH2 solutions
- RadioPharma use show in TKI's afternoon presentation

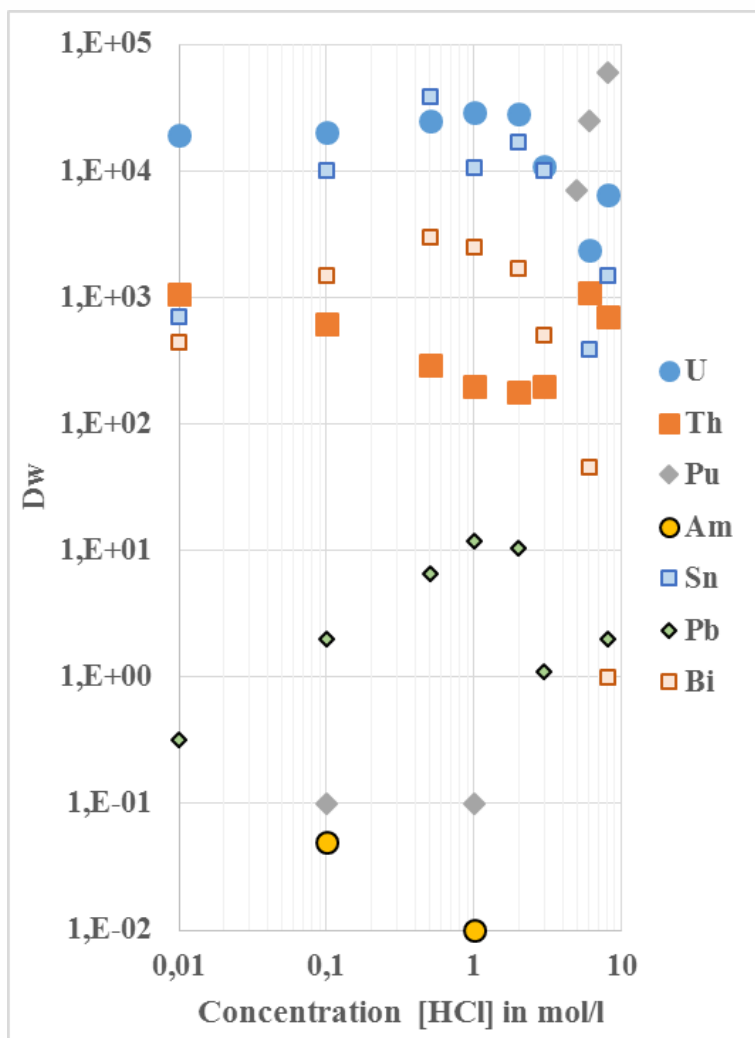


Braun T. et al., Extraction Chromatography, Journal of Chromatography Library (vol2), Elsevier Scientific Publishing Company – 1975, pp.279-303

TK200 Resin – Dw studies HNO_3



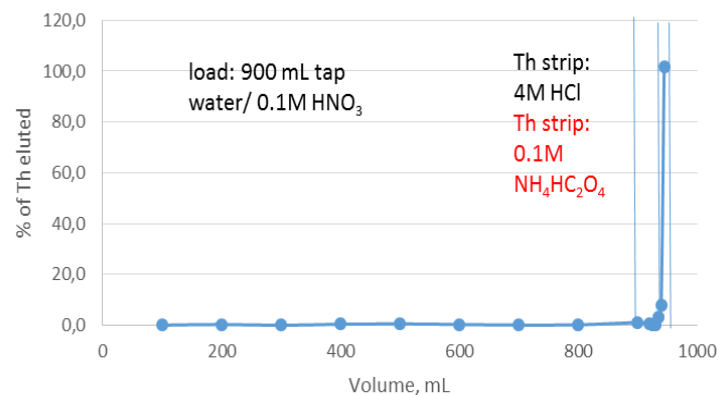
- Retention of Am < 0,1M HNO_3 ;
- U/Th/Pu uptake over the whole acidity range studied;
- High uptake of Bi from 0,01 – 2M HNO_3 => possibility to separate from Pb in case of MS measurement;
- Uptake of Sn from 0,1 – 10M HNO_3 (alternative to TBP Resin).



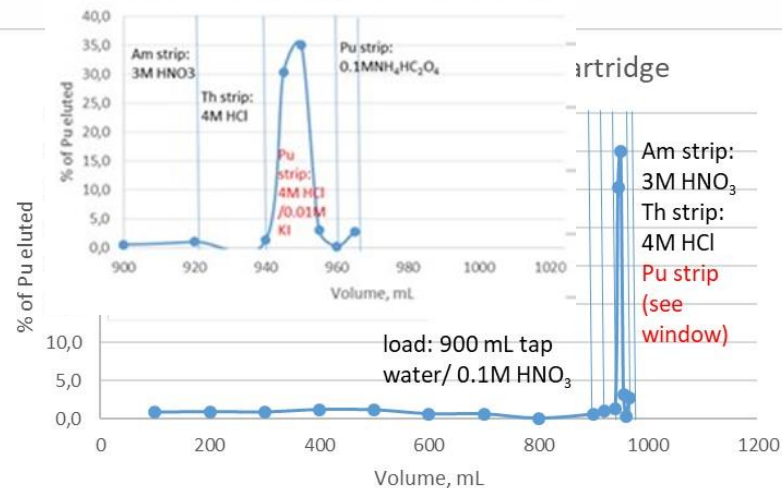
- No retention of Am ;
- U/Th uptake over the whole acidity range studied;
- Pu uptake from 3-10M HCl – no retention below 3M HCl;
- High uptake of Bi from 0,01 – 3M HCl => possibility to separate from Pb in case of MS measurement;
- High uptake of Sn over the whole acidity range studied (alternative to TBP Resin)

Actinides on TK200 – Application

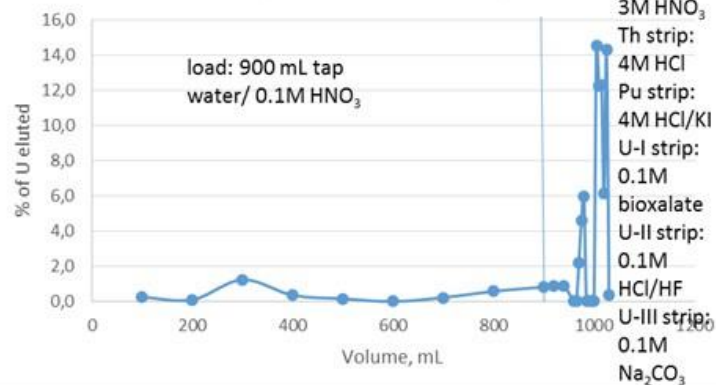
Elution of Th from 1g TK200 resin cartridge



Elution of Pu from 1g TK200 resin cartridge



Elution of U from 1g TK200 resin cartridge

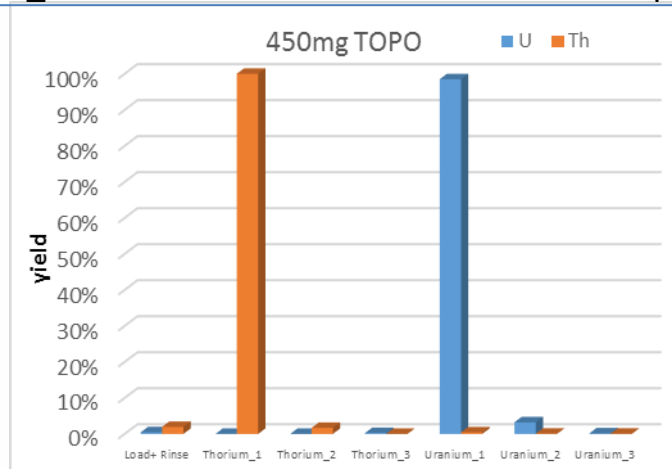


- Method development on-going
- Aim, direct load of actinides from acidified water
- Sequential separation

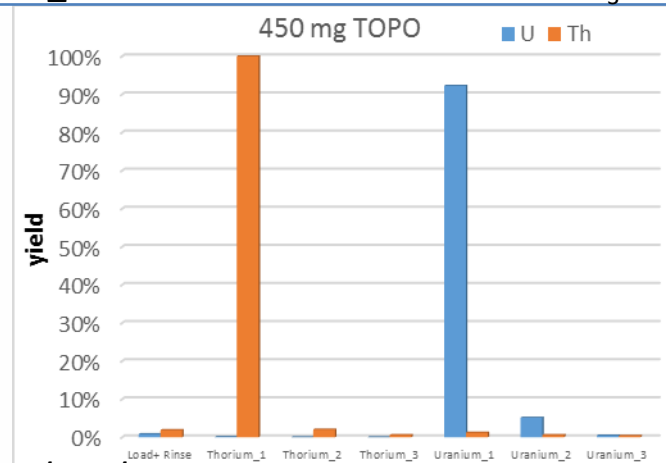
Data from N. Vajda - Radanal

TK200 Resin – Elution studies for U/Th separation from acidic solutions

Load+ Rinse	5mL Load 3 M HNO ₃ + 5mL 3 M HNO ₃
Th_1	10 mL 0.1 M HCl-0.1 M oxalic acid
Th_2	5 mL 0.1 M HCl-0.1 M oxalic acid
Th_3	5 mL 0.1 M HCl-0.1 M oxalic acid
U_1	10 mL 0.1 M Ammoniumoxalate pH 9
U_2	5 mL 0.1 M Ammoniumoxalate pH 9
U_3	5 mL 0.1 M Ammoniumoxalate pH 9



Load+ Rinse	5mL Load 3 m HNO ₃ + 5mL 3 m HNO ₃
Th_1	10mL 0.1 m HCl-0.1 m oxalic acid
Th_2	5 mL 0.1 m HCl-0.1 m oxalic acid
Th_3	5 mL 0.1 m HCl-0.1 m oxalic acid
U_1	10 mL 0.1 m NaHCO ₃
U_2	5 mL 0.1 m NaHCO ₃
U_3	5 mL 0.1 m NaHCO ₃



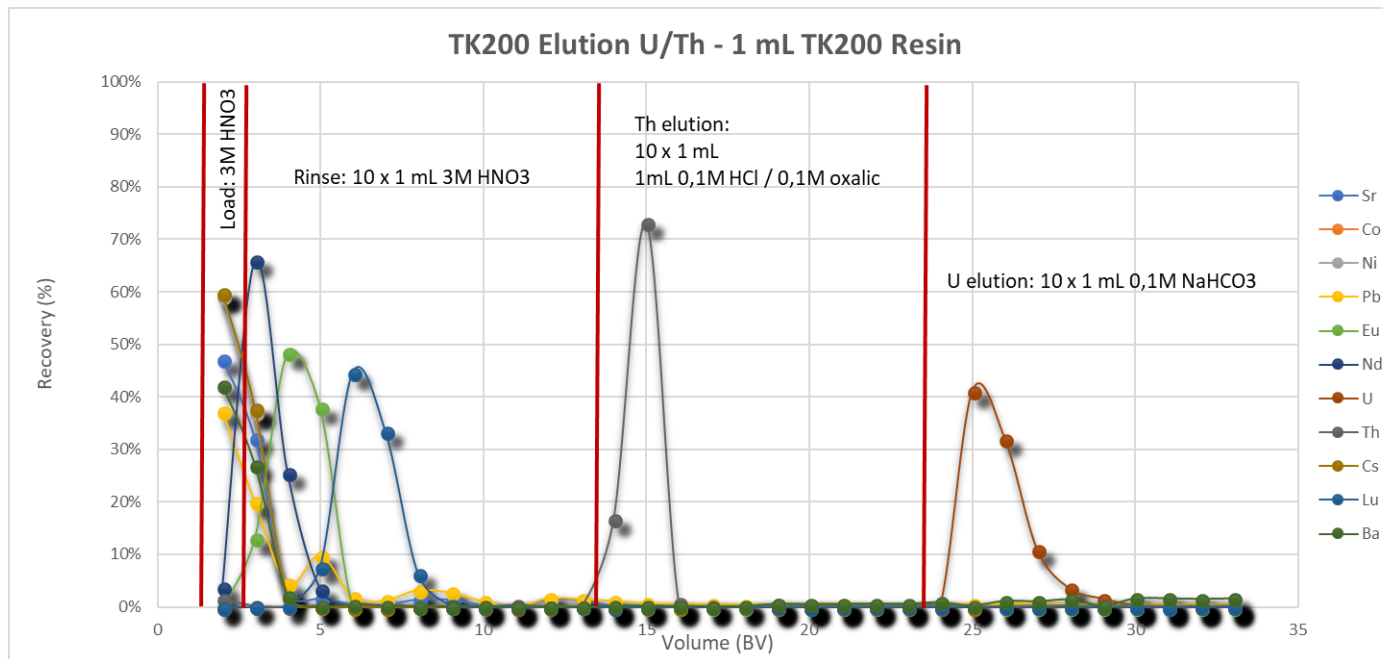
Data from N. Vajda - Radanal

Th selectively separated from U and recovered quantitatively

U quantitatively recovered with 15mL of various solutions depending on needs

=> Good alternative resin to UTEVA resin

U/Th separation on TK200

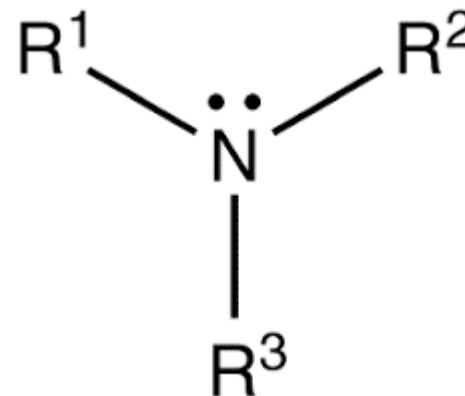


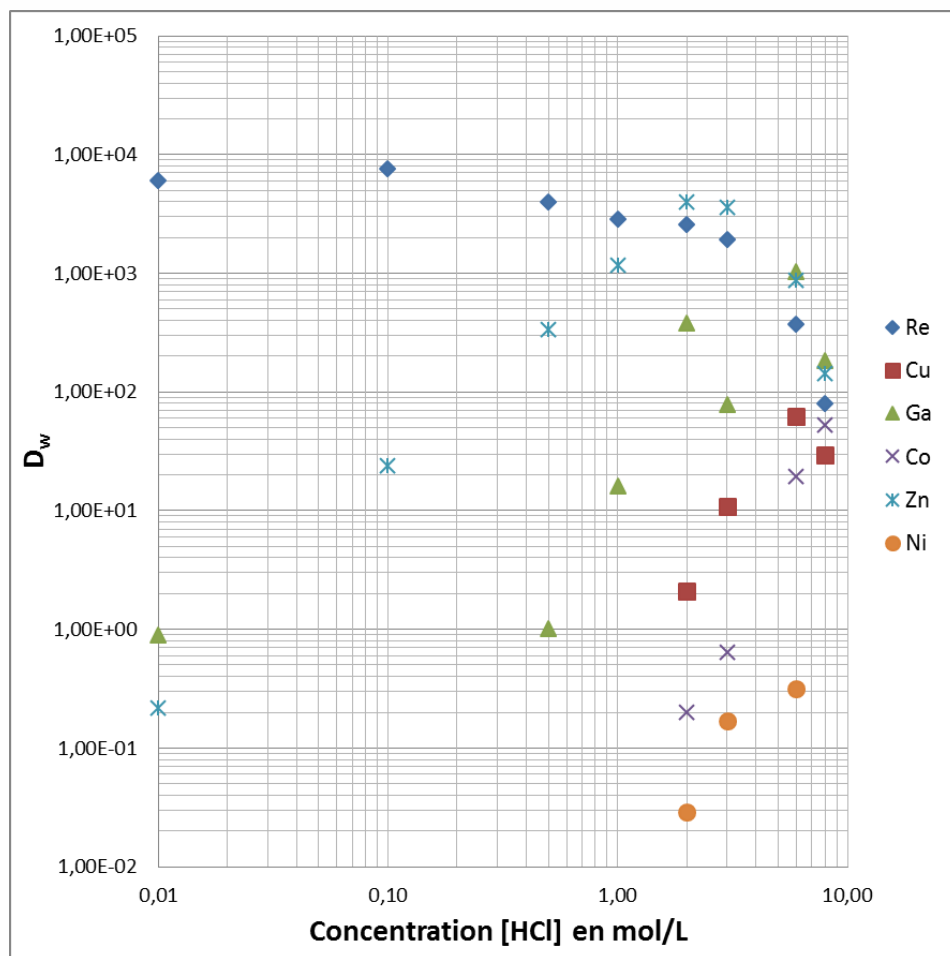
- Load: 3M HNO₃ or $\geq 1\text{L pH}2$ (HNO₃)
- Very clean U/Th separation

- Preconcentration of actinides from pH1-2 solutions => medium usually used to preserve samples for storage and prior to analysis
- Alternative to UTEVA Resin for Th/U separation from acidic conditions
- Possibility to extract/concentrate Sn and Cd in HCl and elute in low HNO_3 concentration.
- Zr/Hf are well extracted in HCl (1-10M) and HNO_3 (whole studied range)

- Tc-99 (difficult to measure – DTM Radionuclide) – 100% beta emitter
- Interest in decommissioning and radioactive waste management and in Nuclear medicine
- TEVA resin allows for Tc separation but quantitative elution needs highly acidic medium
- 2 new resins developed for load with both acidic or alkaline media and specific elution in slightly alkaline or water

- Based on tertiary amine (weak Anion Exchanger) impregnated on inert support
- In acidic medium,
 $R_1(R_2)(R_3)-N^+H Cl^-$
Exchange of the counter anion

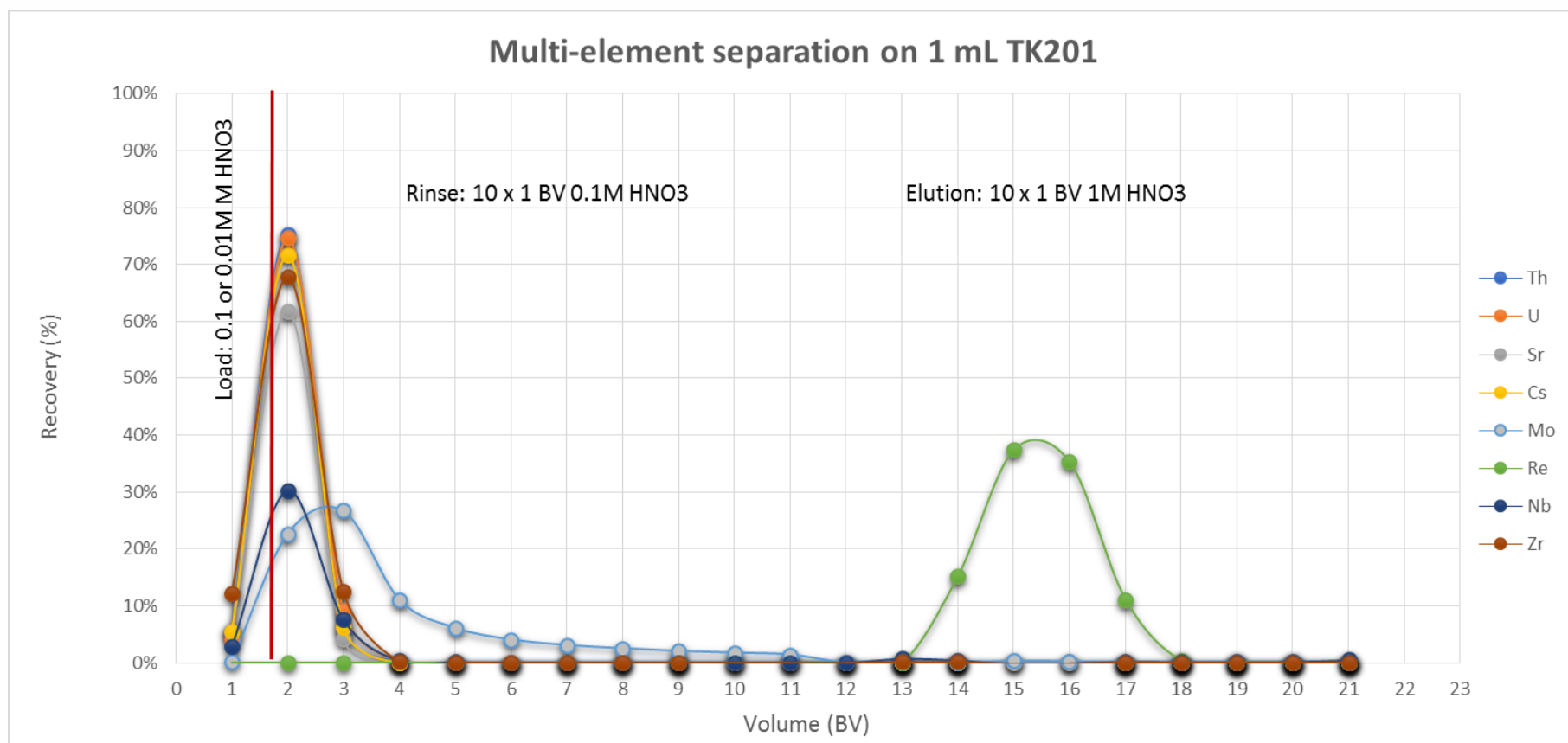




- Re uptake between pH 2 – 5M HCl
- In HNO₃ medium, Re fixed at pH 1-2

TK201 Resin – Elution curve

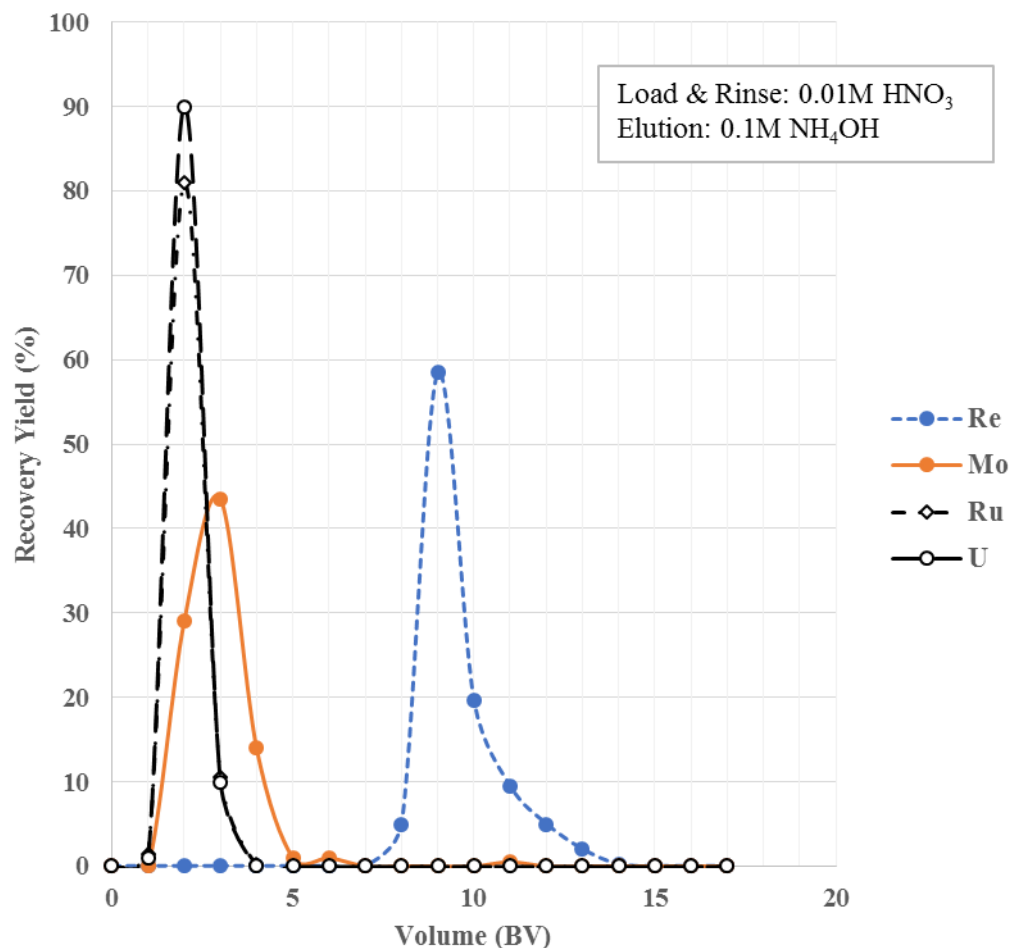
- Load of sample at pH 1-2 to fix Re(Tc)
- Elution of Re @ 1M HNO₃



TK201 Resin – Elution curve

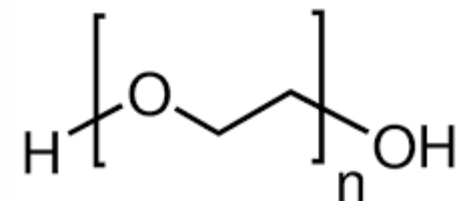
Separation Re(Tc)/Mo

TK201 Resin



- Clean separation of Re in 6BV
0.1M NH₄OH
- Elution of Re also possible with
1M HNO₃
- No retention of other elements
studied

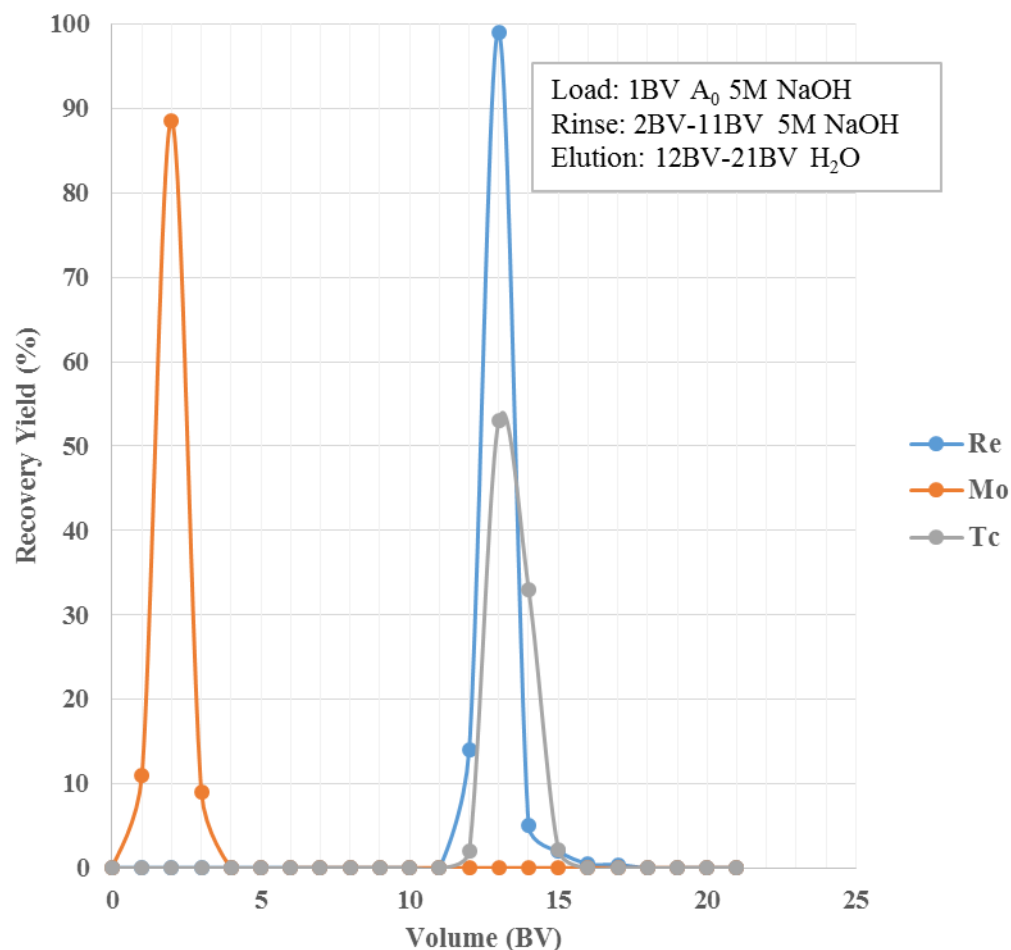
- Based on Polyethylene Glycol (PEG) grafted on inert support



- Exchange from rich PEG phase to poor PEG phase based on
 - salting out effect due to 5M NaOH medium in our case
 - ability of salt to easily dissociates and break hydrogen bonds

TK202 Resin – Elution curves

TK202 Resin



- Tests show Re-Tc have similar behaviour in tested conditions
- Clean separation of Re-Tc and recovery in 5BV H₂O

- 2 resins allowing for the separation of Tc either from starting acidic or alkaline media (liquid wastes/ alkaline dissolutions of concretes for analysis in decommissioning)
- Elution with much less acidic solution (compared to TEVA) or water in case of TK202 Resin
- Further testing going-on with real samples for resins robustness for Tc/Mo separation
- Use in MS for isobars elimination prior to analysis

- Two resins:
 - AMP-PAN (acidic samples)
 - KNiFC-PAN (slightly acidic to neutral samples)
 - 70 – 85% active component, fast kinetics
- Both resins efficiently remove Cs, very difficult to regenerate
- Typical applications:
 - Low level Cs-134/7 determination of water samples by gamma spectrometry => bulk, columns, cartridges
 - Removal of high levels of Cs-137 before MS => safeguards
 - Cs-137 decontamination of contaminated effluents => bulk, large quantities and particle sizes



Cs measurements in Seawater [8][9]

- Procedure:
 - Seawater Sample volumes: 100L,
 - Acidified (pH 1-2) and raw samples,
 - Column bed 25ml of AMP-PAN or KNiFC-PAN,
 - Flowrate: maximum at 300ml.min⁻¹,
 - Gamma spectrometry measurement

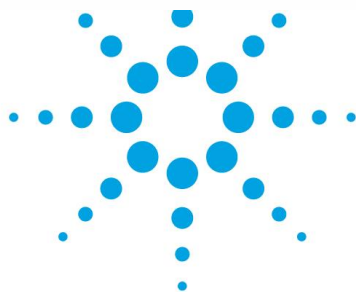
➤ Results: Yield > 90% Cs retention



[8] Pike et al., Extraction of Cesium from Seawater off Japan using AMP-PAN Resin and Quantification via Gamma Spectrometry and Inductively Coupled Mass Spectrometry, J. Radioanal. Nucl. Chem, DOI 10.1007/s10967-012-2014-5, 2012

[9] Kamenik J. et al., Fast Concentration of Dissolved forms of Cesium Radioisotopes from Large Seawater Samples, J. Radioanal. Nucl. Chem, DOI 10.1007/s10967-012-207-4, 2012

- Development for measurement of ^{90}Sr and ^{226}Ra by Russel and Van Es from NPL



Rapid Analysis of Radium-226 in Water Samples by ICP-QQQ

Application Note
Nuclear, environmental

Authors

Ben Russell¹, Elsje May van Es^{1,2},
Glenn Woods³, David Read^{1,2}

1. National Physical Laboratory,
Teddington, UK

2. Chemistry Department, University
of Surrey, Guildford, Surrey, UK



https://www.agilent.com/cs/library/applications/8800_ICP-MS_5991-8324EN_radium_analysis.pdf

Results and applications to be presented by

- Peter Ivanov
- Chris Gilligan

Later during this meeting



Thank you for your attention!

New website @ www.triskem.com

Radiopharmacy
and
Nuclear Medicine

Environment and
Bioassay

Geochemistry
and
Metals Separation

Decommissioning

UGM Cambridge - 21/09/18