

# Rapid determination of Pb-210 and Sr-90 in water samples using new crown-ether based extraction chromatographic resins

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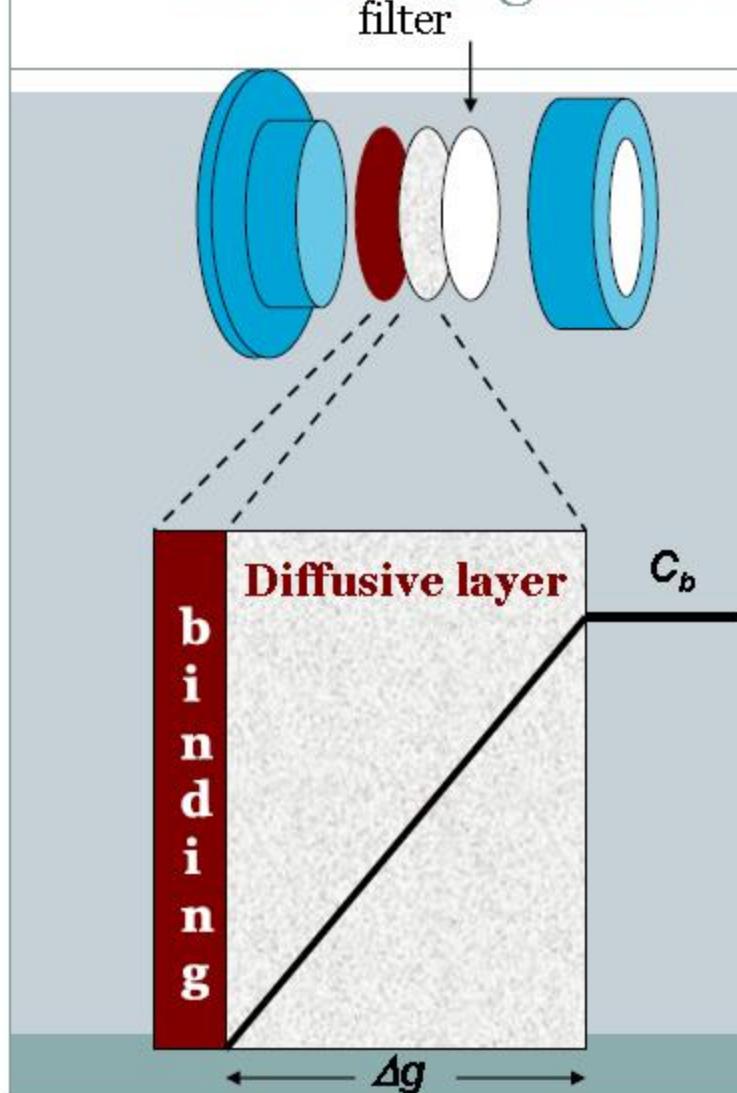
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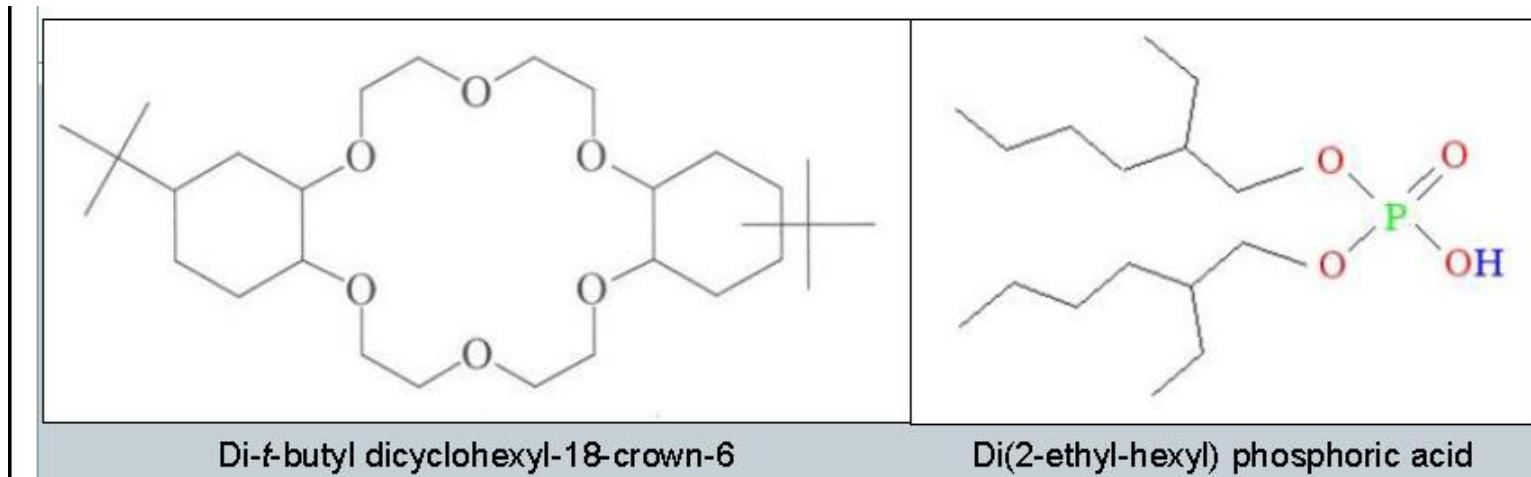
# Context

- On-going work
- Original project: Rapid extraction and separation of Sr from water samples (pH5–8)
- « Passive sampling »
  - Use in DGT (Diffusive gradients in thin-films) units
  - Weakly bound/complexed species (« bioavailability »)
  - Technique also used in NORM monitoring
    - Ra-226 via MnO<sub>2</sub> (resin and Ra Nucfilm discs), U
- Rapid method
  - Concentration and separation on same resin/column
  - Load in batch, disc or column mode

# Diffusive gradients in thin-films (DGT)

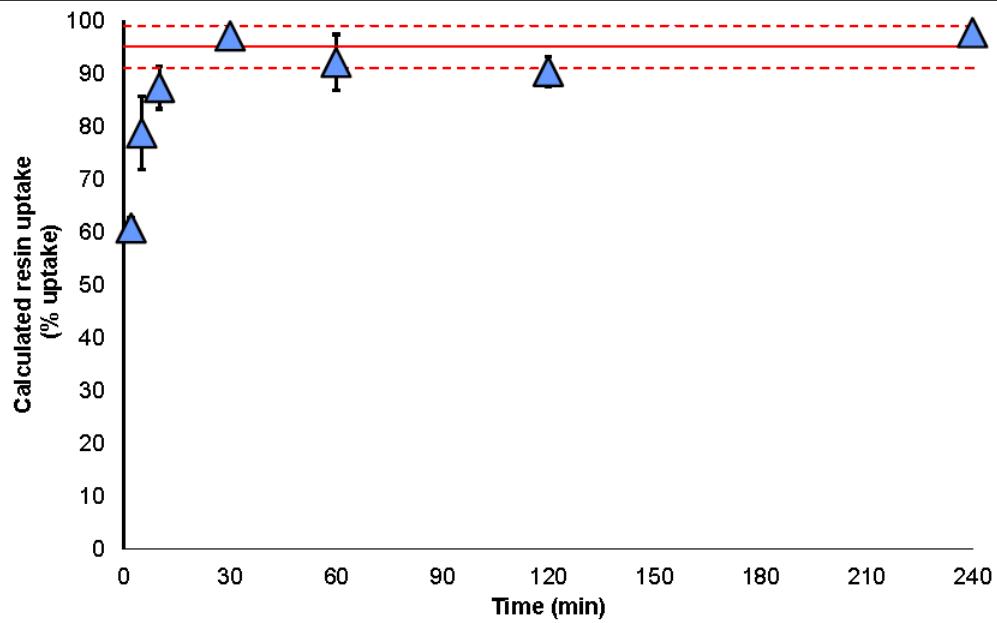


# TK100 Resin



- Keep Sr Resin selectivity (crown ether), increase pH range
- 1<sup>st</sup> approach: Replace 1-Octanol by HDEHP
- Screening of several test resins (varying HDEHP contents)
- Characterisation of best suited resin prototype
- Elution studies and first tests

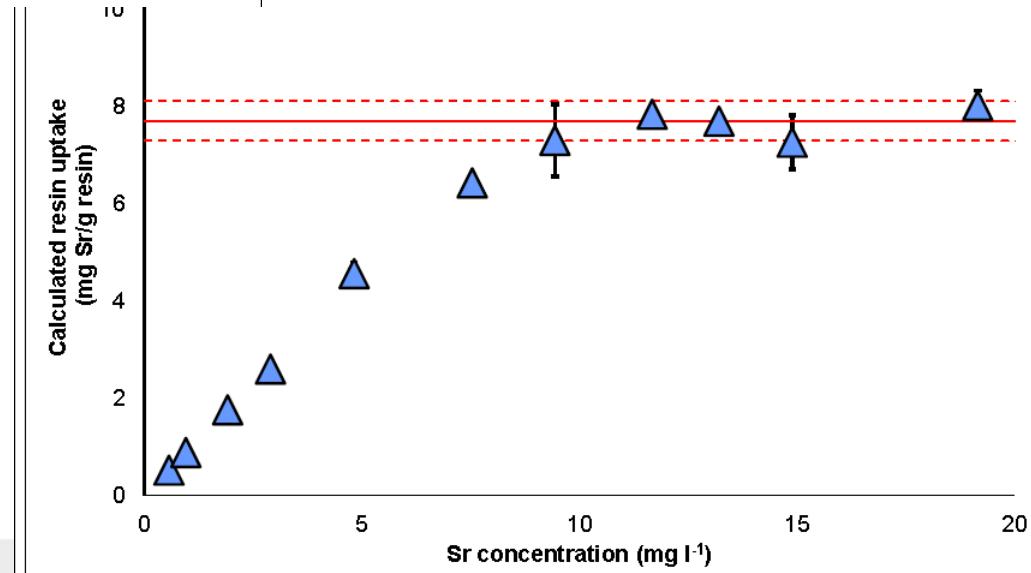
## Uptake kinetics



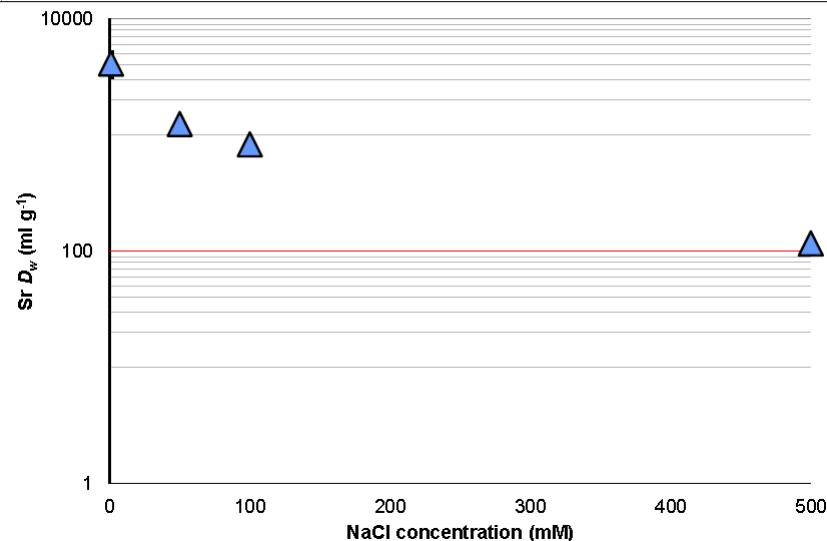
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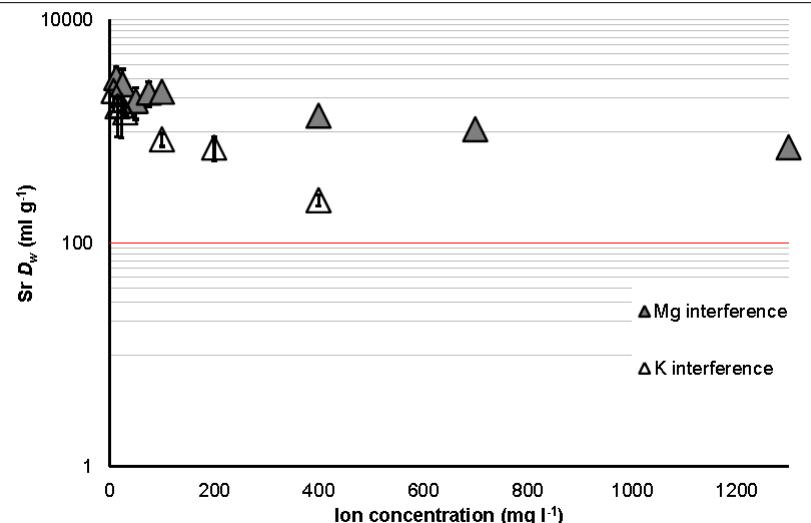
## Capacity



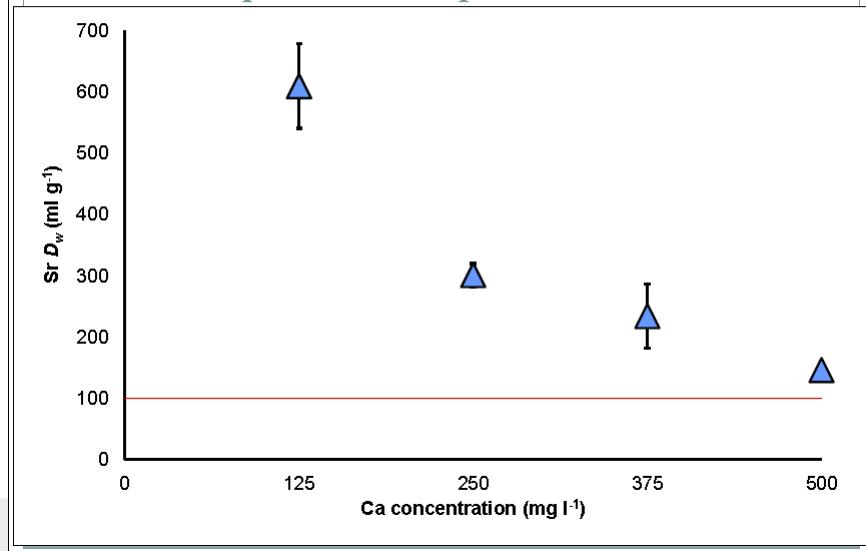
## Sr uptake and ionic strength



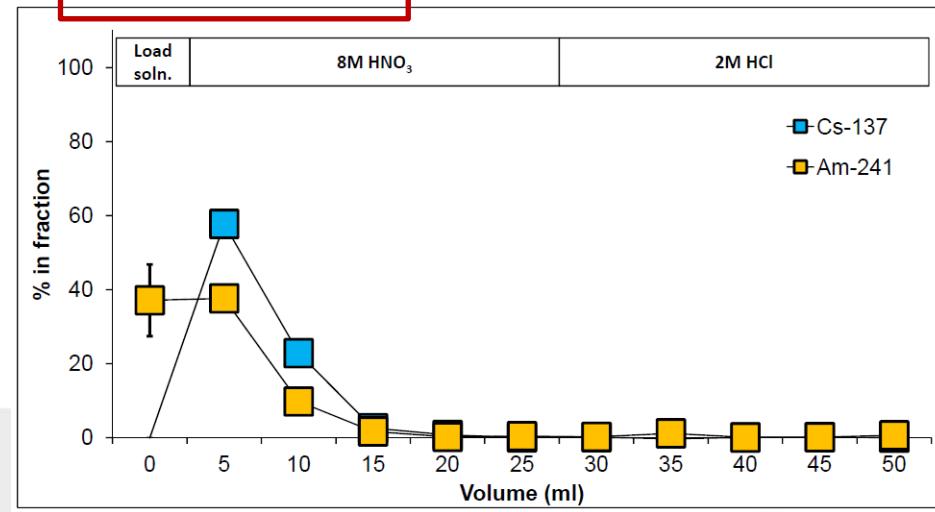
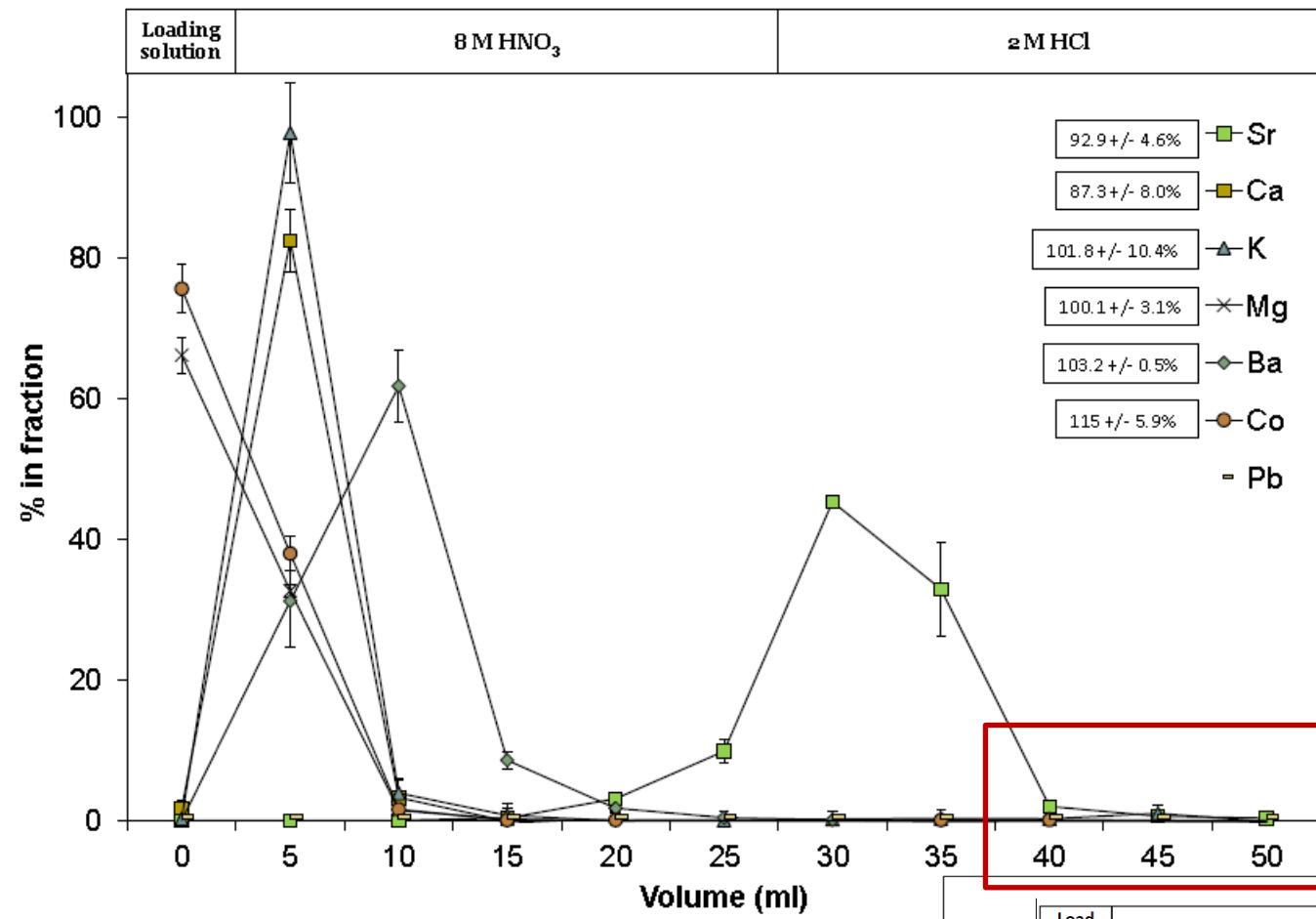
## Sr uptake in the presence of Mg and K



## Sr uptake in the presence of Ca



- pH 7
- Batch experiments



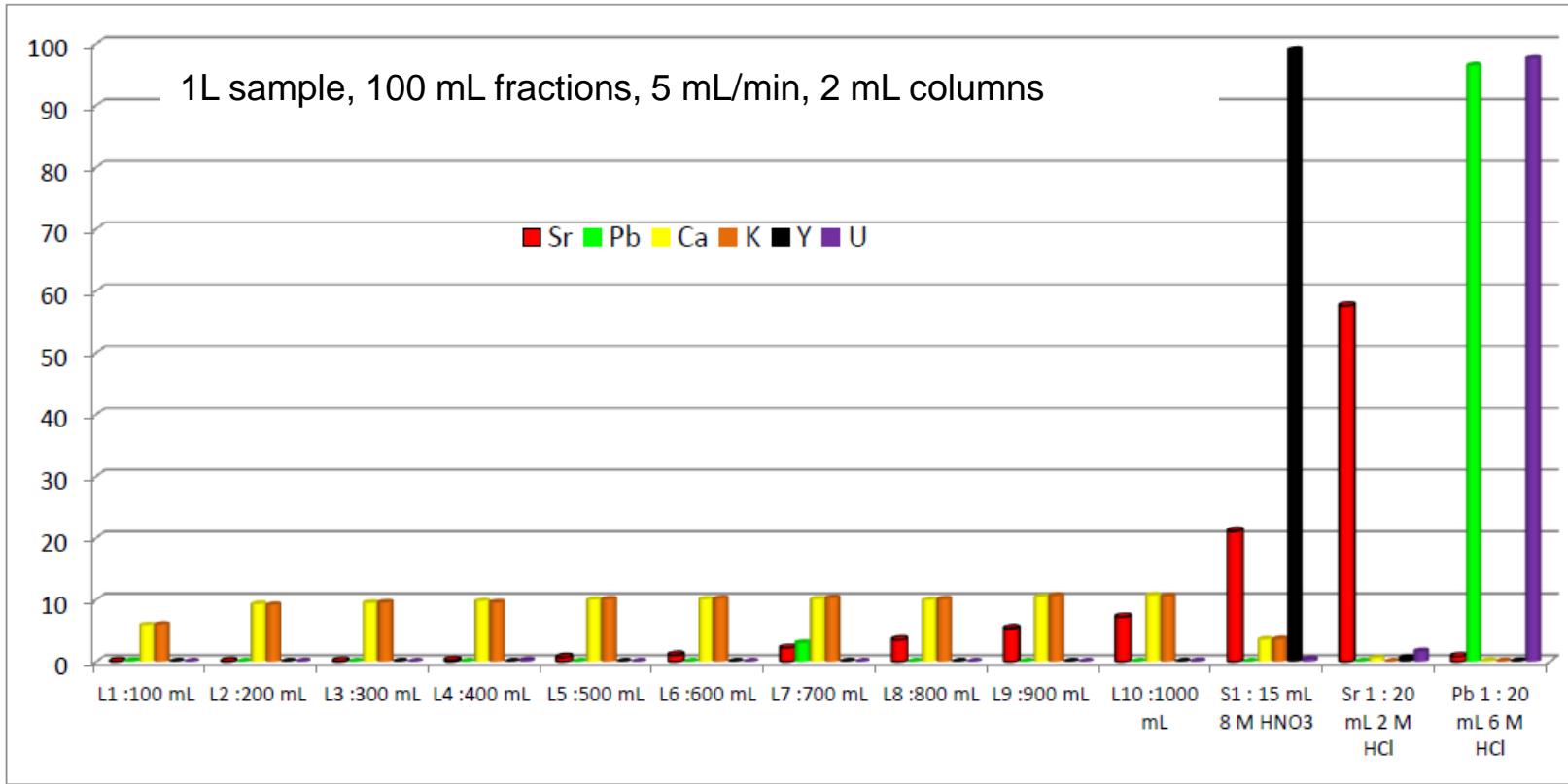
# Batch experiment

- 0.5 g of resin added to 1 litre of DI water spiked with  $^{90}\text{Sr}$ , shaken for 1 hour
- Supernatant and resin separated, resin loaded onto a column with a 0.4 g ‘guard layer’ of fresh resin to lower Sr breakthrough
- Column washed with 15 ml 8M  $\text{HNO}_3$  and 20 ml 2M HCl
- Fractions collected and counted by LSC
- Sr yield ~73%

# Application tests

- Elution study 1L sample (column experiment)
  - pH7, 1 mg Sr, 100 mg Ca, 5 mg K, 0.1 mg Pb, Y, U per sample
  - 1L samples, 100 mL aliquots
  - 2 mL columns (650 mg resin)
  - Vacuum supported separation, 5 mL/min
  - Incl. Pb elution step (6M HCl)
  - ICP-MS measurement of effluents

# Column breakthrough study – direct load



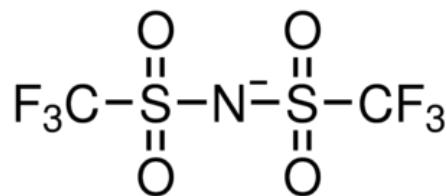
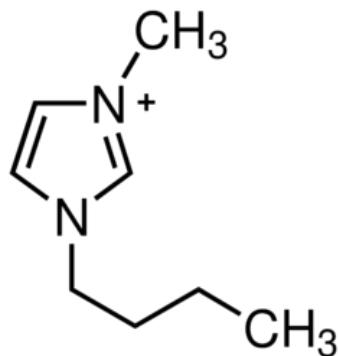
- K and Ca not retained
- Sr breakthrough starts at 600 mL
- Pb fraction:
  - high Pb yield
  - also containing U

# Sr Elution study

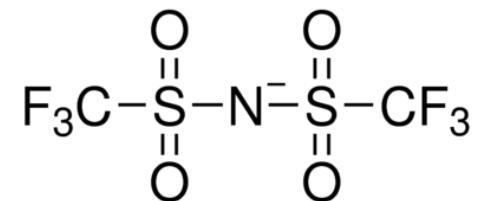
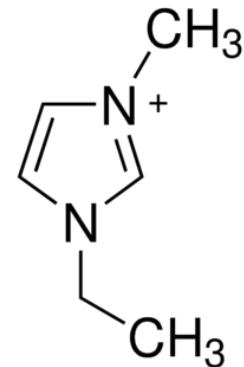
- Elution studies
- 250 mL and 500 mL spiked tap water samples (pH 2, each N=3)
- Load at 5 – 10 mL/min
- Rinse:
  - 5 mL deion. water
  - 15 mL 8M HNO<sub>3</sub>
- Elution with 20 mL 2M HCl
- 250 mL: 95,2 +/- 2,5
- 500 mL: 88,2 +/- 4,3
- Automatisation tests on-going (FZ Jülich)

# Improvement of selectivity

- Use of room temperature liquids (RTILs) instead of HDEHP

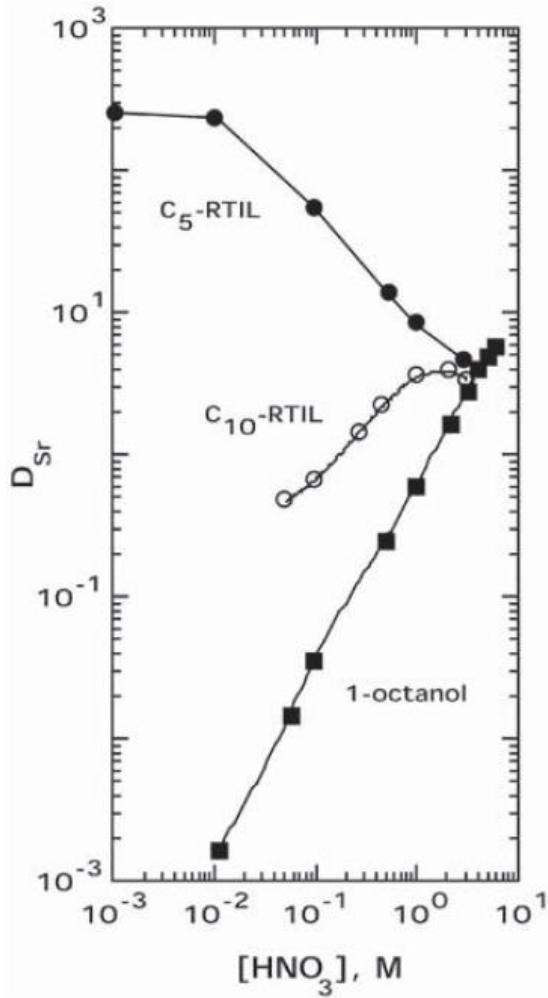


1-Butyl-3-methylimidazolium  
bis(trifluoromethylsulfonyl)imide (**BMIM**)



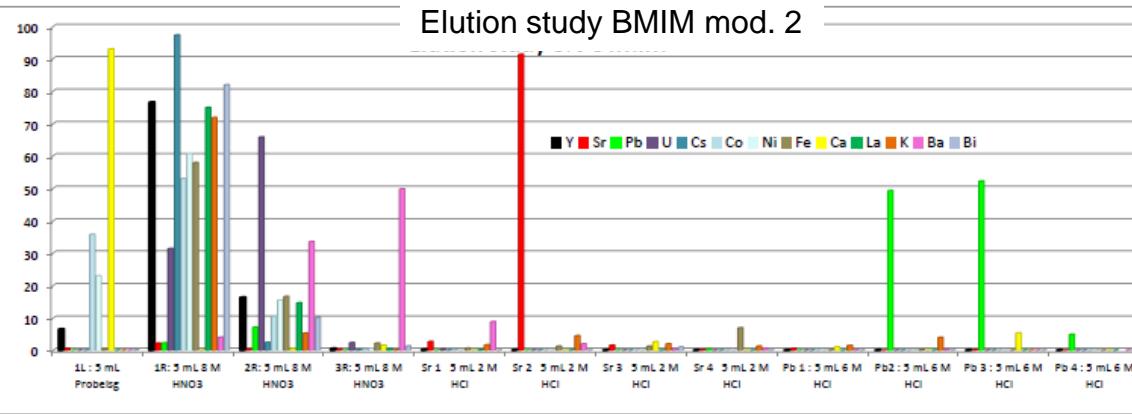
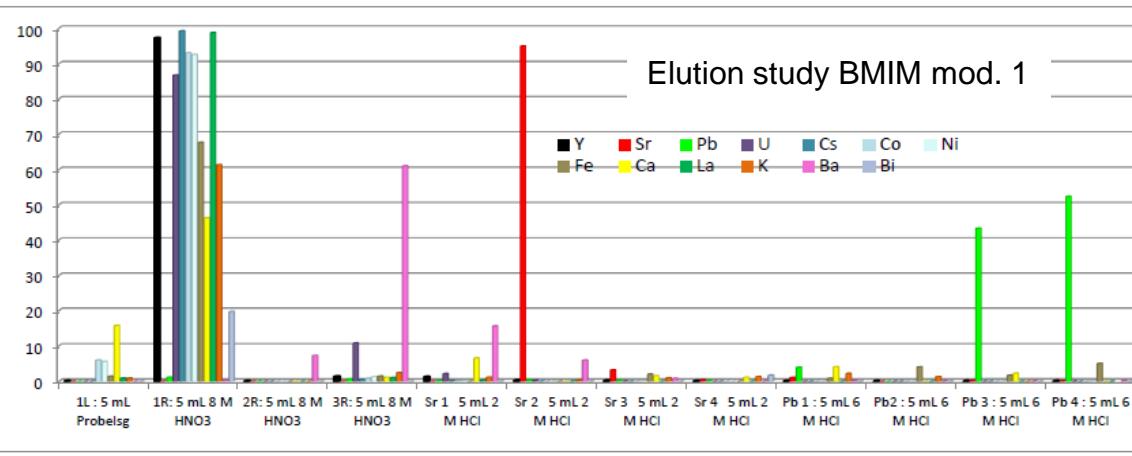
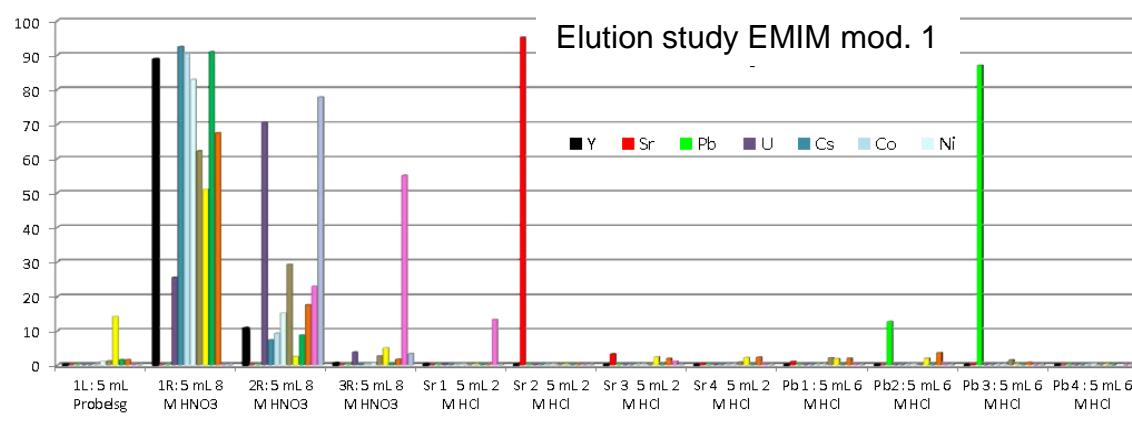
1-Ethyl-3-methylimidazolium  
bis(trifluoromethylsulfonyl)imide (**EMIM**)

# Influence of Room Temperature Ionic Liquids (RTILs) on Sr extraction by crown-ethers



- Dietz et al. 2003:
- Short-chain RTIL:  
ion exchange mechanism
- Long-chain RTIL and octanol:  
extraction of neutral species

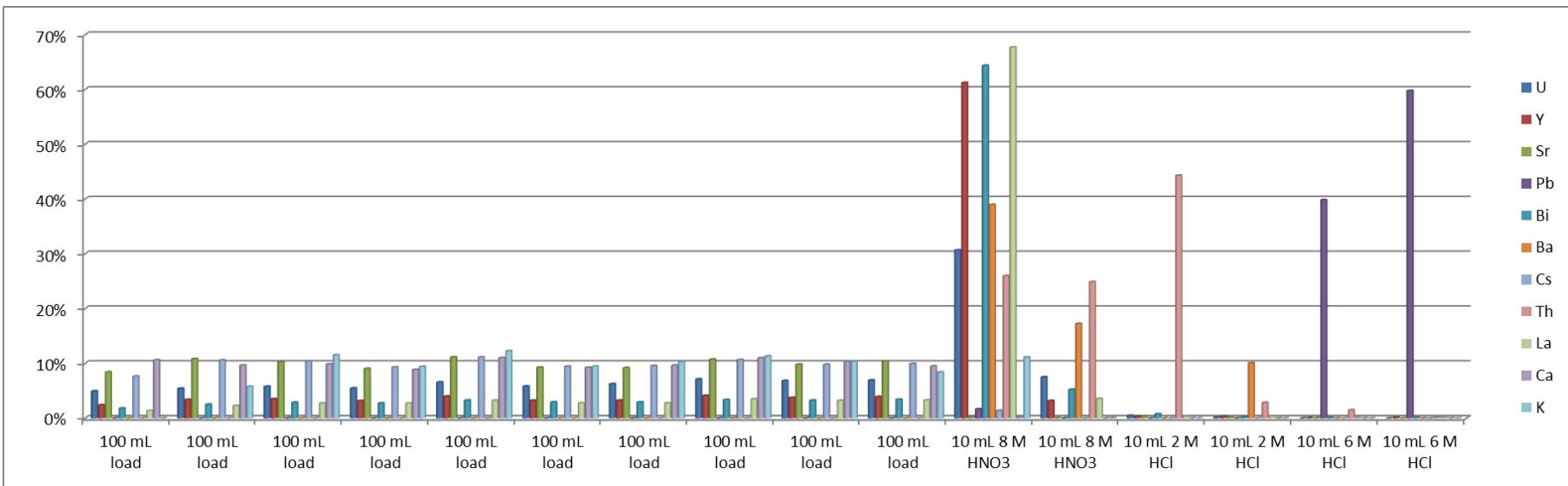
Fig. 1 Nitric acid dependency of  $D_{\text{Sr}}$  for DCH18C6 (0.10 M) in 1-octanol,  $\text{C}_5\text{mim}^+\text{Tf}_2\text{N}^-$ , and  $\text{C}_{10}\text{mim}^+\text{Tf}_2\text{N}^-$ . ( $T = 23^\circ\text{C}$ ).



- Load: of multi-element solution at pH 7
- Rinse with 3x5 mL 8M HNO<sub>3</sub>,
- Sr elution with 4x5 mL 2M HCl
- Pb elution with 4x5 mL 6M HCl
- Similar elution characteristics
- Improved purity of Pb fraction
- Breakthrough experiments on-going (larger volumes)

# Elution study 1L sample

- 1L sample, pH 2, multi-element spike
- Load in 100 mL steps (all eluates collected and analysed), 5 – 10 mL/min
- Separation as described before



- No Pb breakthrough
- Most other elements not retained or removed with 8M HNO<sub>3</sub>
- Th and trace Ba removed in 2M HCl
- Pb quantitatively eluted with 20 mL 6M HCl
- Automatization tests on-going (FZ Jülich)

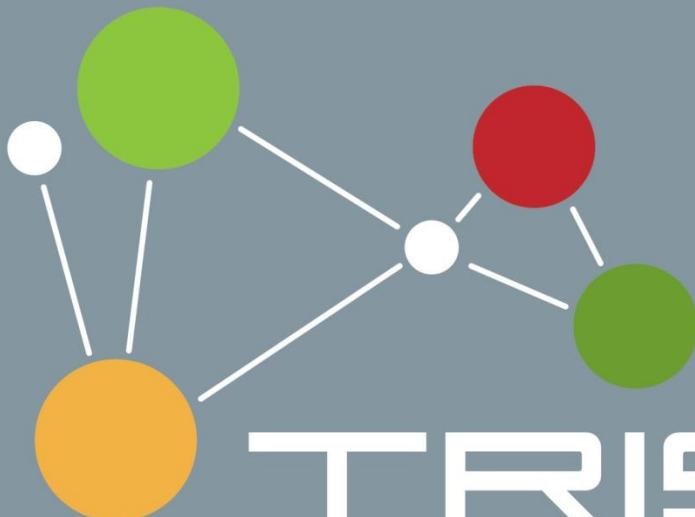
# Extractive disc

- Aim: Direct LSC measurement of discs
  - Sample load and rinse (incl. additional rinse with water to remove trace acid)
  - Drying with vacuum
  - Transfer disc in LSC vial
  - Addition of 2 mL water and 16 mL LSC cocktail (ProSafe HC+ )
- First tests on LSC counting
  - Low increase in blank count rate
  - Count rate is not depending on orientation of disc in vial:
    - 3 samples prepared (known Pb-210 activity on disc)
    - Each counted four times (90° turns after each measurement)
    - %RSD of count rate < 1% (N=12, k=1, at 100 Bq level)

# Conclusion

- On-going work
- Crown-ether based resins
  - Modification of Sr resin with HDEHP and RTILs results in extended uptake pH range (pH 8 to high nitric acid) for Sr and Pb
- High Pb uptake
- Pb separation possible
  - U/Pb separation to be addressed for HDEHP modified resin
  - Use of RTIL increases Pb selectivity
- Tests in column and disc form
- Automated separation currently being tested
- Discs for increased flow rates
  - Direct LSC count of discs: no dependence on disc orientation
  - Rinsing steps be optimized for discs
- DGT tests

# Thank you for your attention!



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