



Philipps



Universität  
Marburg

# Carbon Nanotubes

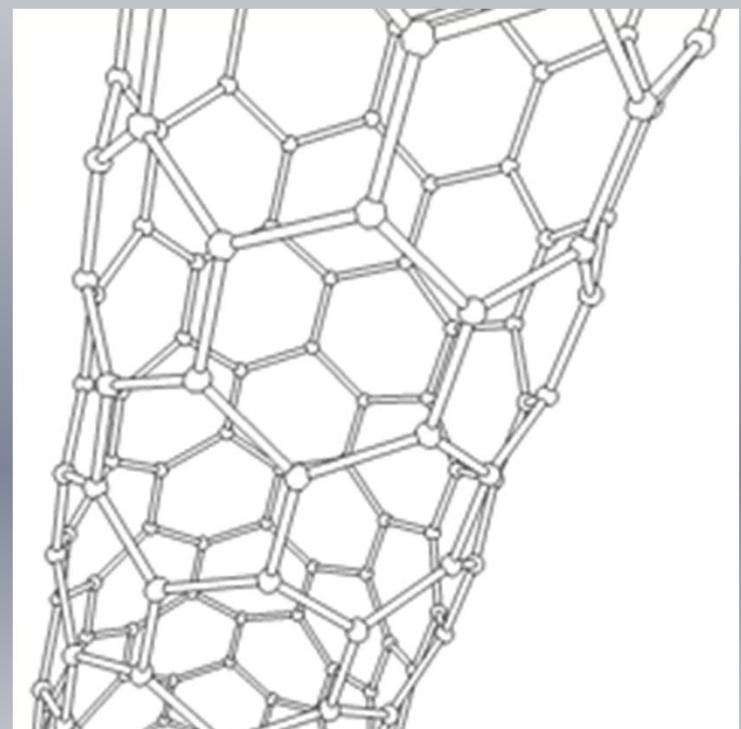
Dirks, Carina.<sup>1</sup>, Happel, S.<sup>2</sup>, Jungclas, H.<sup>1</sup>

[1] Radiochemistry, Department of Chemistry, Philipps-University Marburg, Marburg, Germany

[2] TrisKem International, Bruz, France

# Scope

- What are Carbon Nanotubes?
  - Properties
- Coating of Carbon Nanotubes
- Batch experiments
  - Selectivity
- Column experiments
  - Flow rate
- Summary
- Perspectives



[http://en.wikipedia.org/wiki/File:Kohlenstoffnanoröhre\\_Animation.gif](http://en.wikipedia.org/wiki/File:Kohlenstoffnanoröhre_Animation.gif)

# Dimensions of Nanotubes

- Nano from Greek (nanos = dwarf)
- 1 Nanometer = **1/1 000 000 mm** ≈ 3 Gold atoms

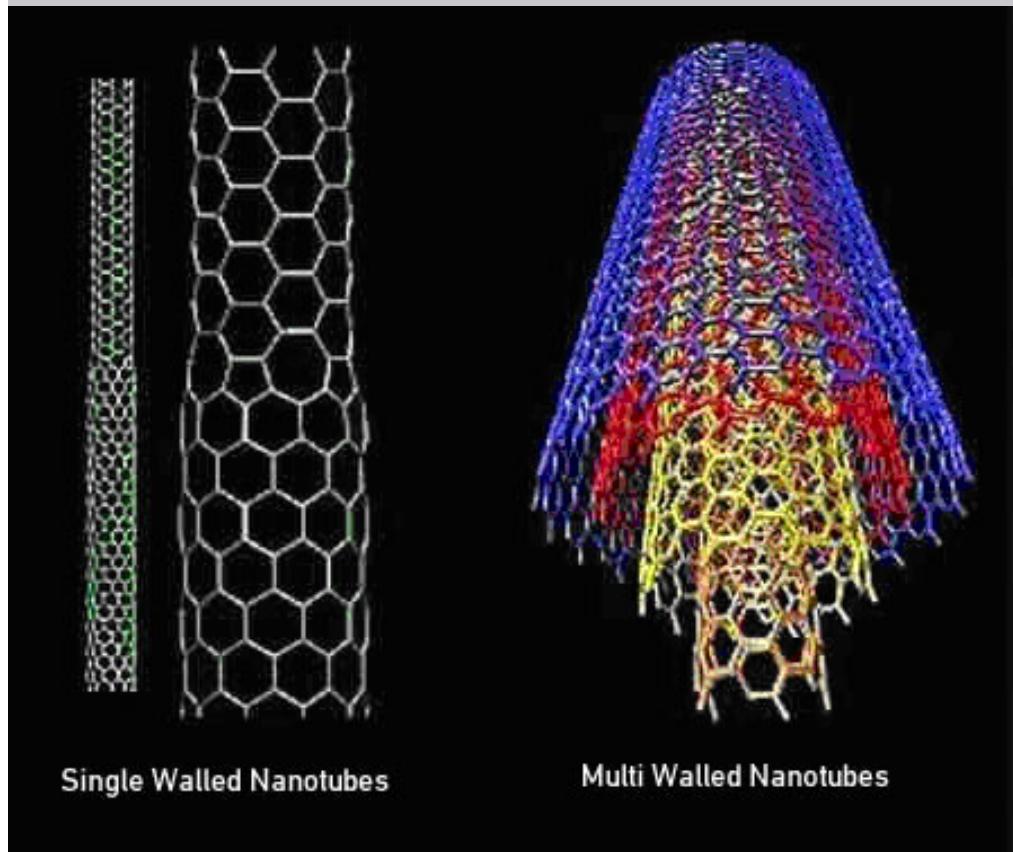


# What are Carbon Nanotubes?

- Another C-modification besides graphene, diamant and fullerene
- $sp^2$  hybridisation
- hexagonal structure → rolled graphene
- Diameter between 0,4 and 100 nm
- Length up to 1 mm (record 20 cm <sup>[a]</sup>)
- Variety of interesting properties
- High specific surface: elevated reactivity

[a] H.W. Zhu, C.L. Xu, D.H. Wu, B.Q. Wei, R. Vajtai und P.M. Ajayan, Science 2002, 296, 884.

# Classification of Nanotubes

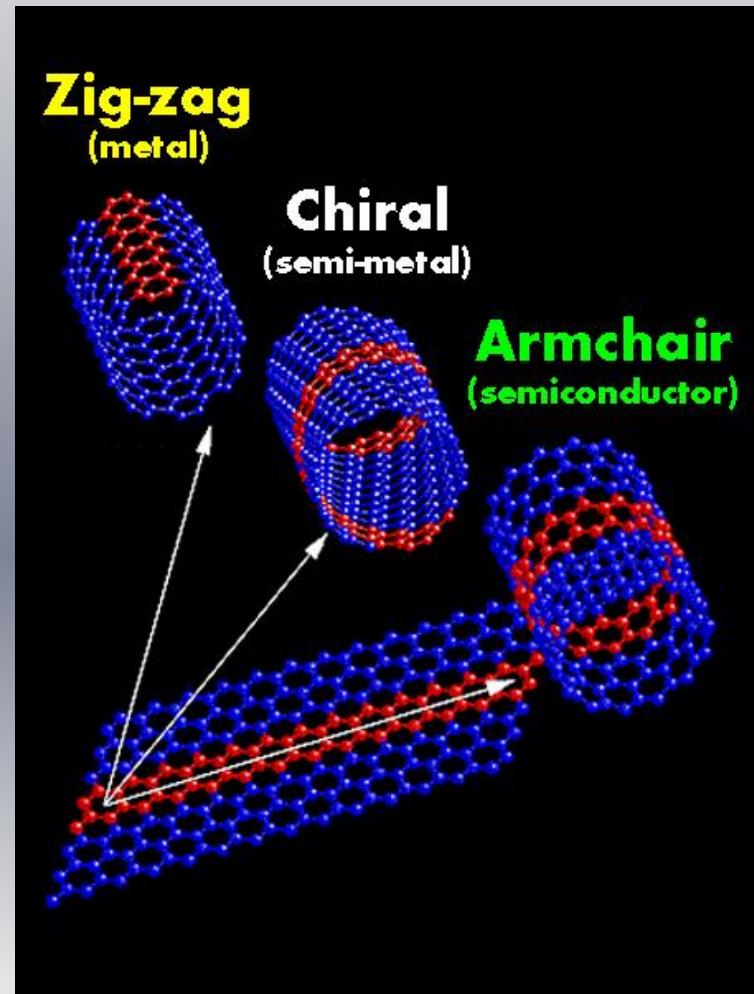
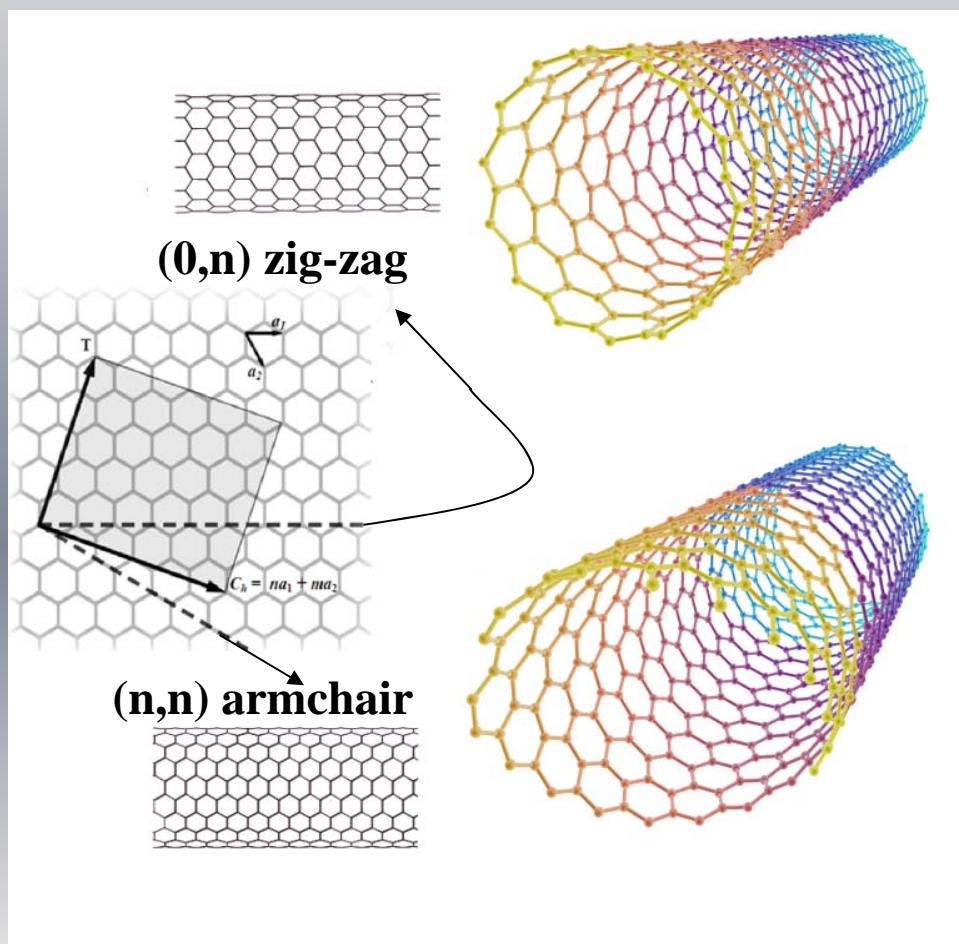


<b>SWCNT</b>	<b>Single-walled-carbon-nanotubes:</b> Rolled Graphene <ul style="list-style-type: none"><li>• 1 nm</li></ul>
Diameter	
<b>MWCNT</b>	<b>Multi-walled-carbon-nanotubes:</b> Concentric SWNCTs
Diameter	<ul style="list-style-type: none"><li>• 5-80 nm</li></ul>

[http://www.tedpella.com/gold\\_html/Nanotubes.htm](http://www.tedpella.com/gold_html/Nanotubes.htm)

<http://coecs.ou.edu/Brian.P.Grady/nanotube.html>

# single-walled Nanotubes



# Properties of the carbon allotropes

Allotrope	Hardness	Tensile strength	Melting behaviour	Conductivity
Coal	+	+	+	No
Graphite	++	++	++++	++++
Diamant	++++	unknown	+++	No
Buckyballs	++++	+++	+	+
Carbon Nanotubes	+++++	++++	++++	+++++

# Carbon Nanotubes

## ➤ Properties:

- Much higher tensile strength than steel, while lighter than steel
- Conductivity similar to Cu
- Thermal conductivity similar to diamant
- Can be conductor or semi-conductor
- **High chemical resistance**

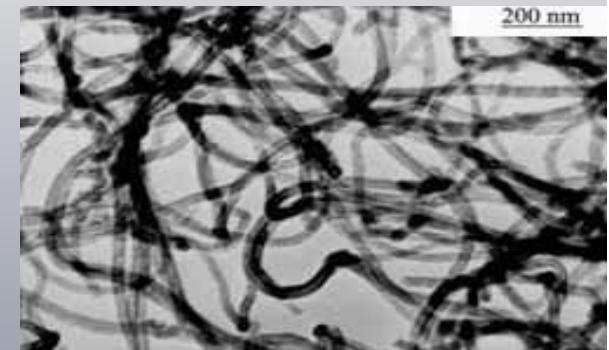
## ➤ For use in columns:

- MWCNT easier to handle and cheaper

# Modified MWCNTs

➤ MWCNT used:

- Industrial production (Chemical Vapor Deposition)
  - Diameter: > 50 nm
  - Length 10-20 mm



Transmission Electron Microscopy (TEM)

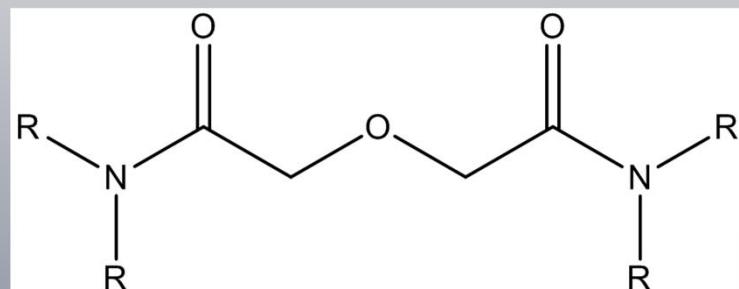
- Coating of MWCNT's with different extractants
- Batch experiments with U-238 and Am-241 in comparison to DGA and TRU Resin
  - TDNC vs. DGA
  - TTNC vs. TRU

# Extractants

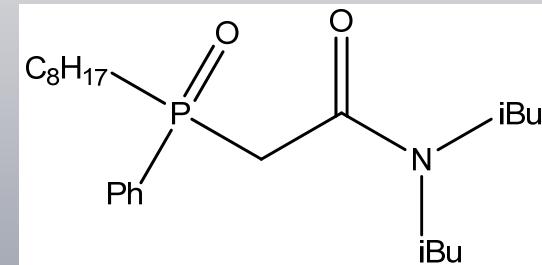
DGA

and

TRU

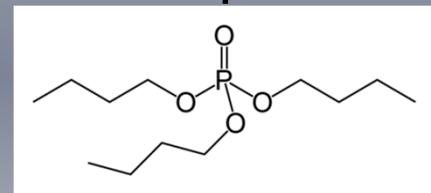


N,N,N',N'-tetra-n-alkyl-3-oxopentandiamide  
(R=C8)

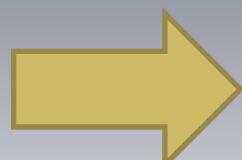


CMPO: (Octyl(phenyl)-N,N-Diisobutyl  
Carbamoylmethyl Phosphinoxide

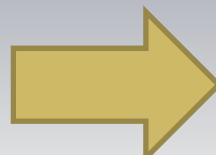
+



TBP: Tributylphosphate



TDNC



TTNC

# General procedure - batch experiments

## $D_w$ (Weight distribution coefficient)

- Weigh 50 mg of the respective resin into a 2 mL Eppendorf cap
- Add 400 µL of the acid to be tested
- Close cap and shake for 30 min (preconditioning)
- Add 1mL of the sample solution  
(e.g.. 1 mL multi-element solution)
- Close cap and shake for 30 min (analyte extraction)
- Centrifuge
- Withdraw 1 mL of the supernatant, Analysis (ICP-MS)
- All  $D_w$  determined in triplicate

$$D_w$$

$$D_w = \frac{N_{A_0} - N_A}{N_A} \times \frac{V}{m_R}$$

- high  $D_w$  = Extraction
- low  $D_w$  = Elution

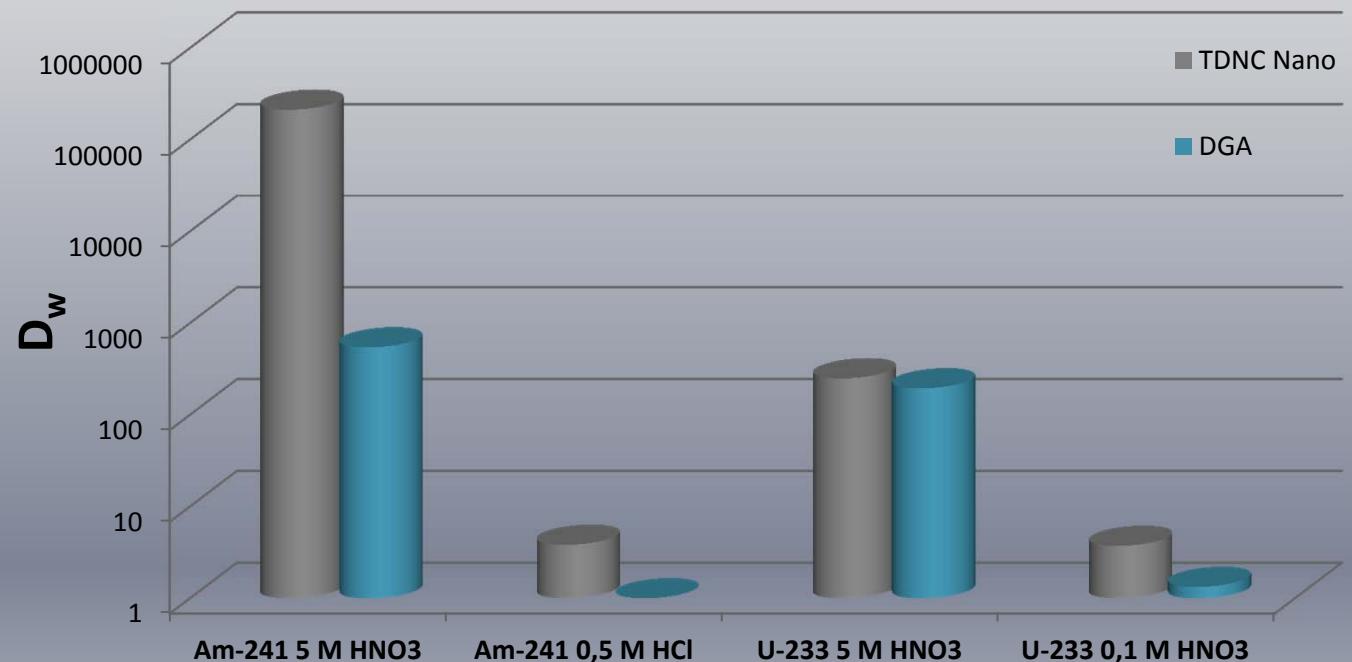
$N_{A_0}$  = Net count rate  $A_o$  sample

$N_A$  = Net count rate sample

$V$  = Volume aq. phase (1,4 mL)

$m_R$  = amount of resin in g

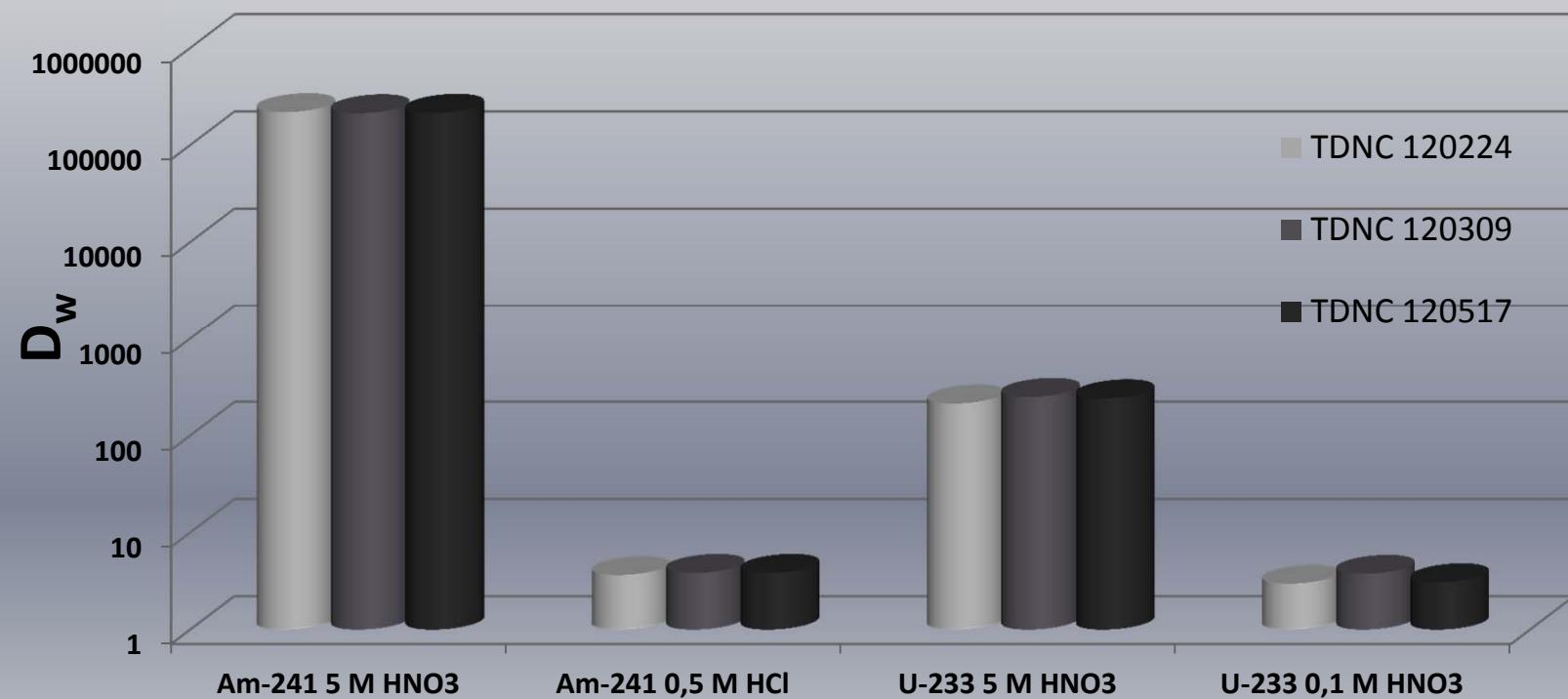
# TDNC vs. DGA



*Comparison coated MWCNT (TDNC) vs DGA Resin*

Very high  $D_w$  Koeffizienten on TDNC especially for Am

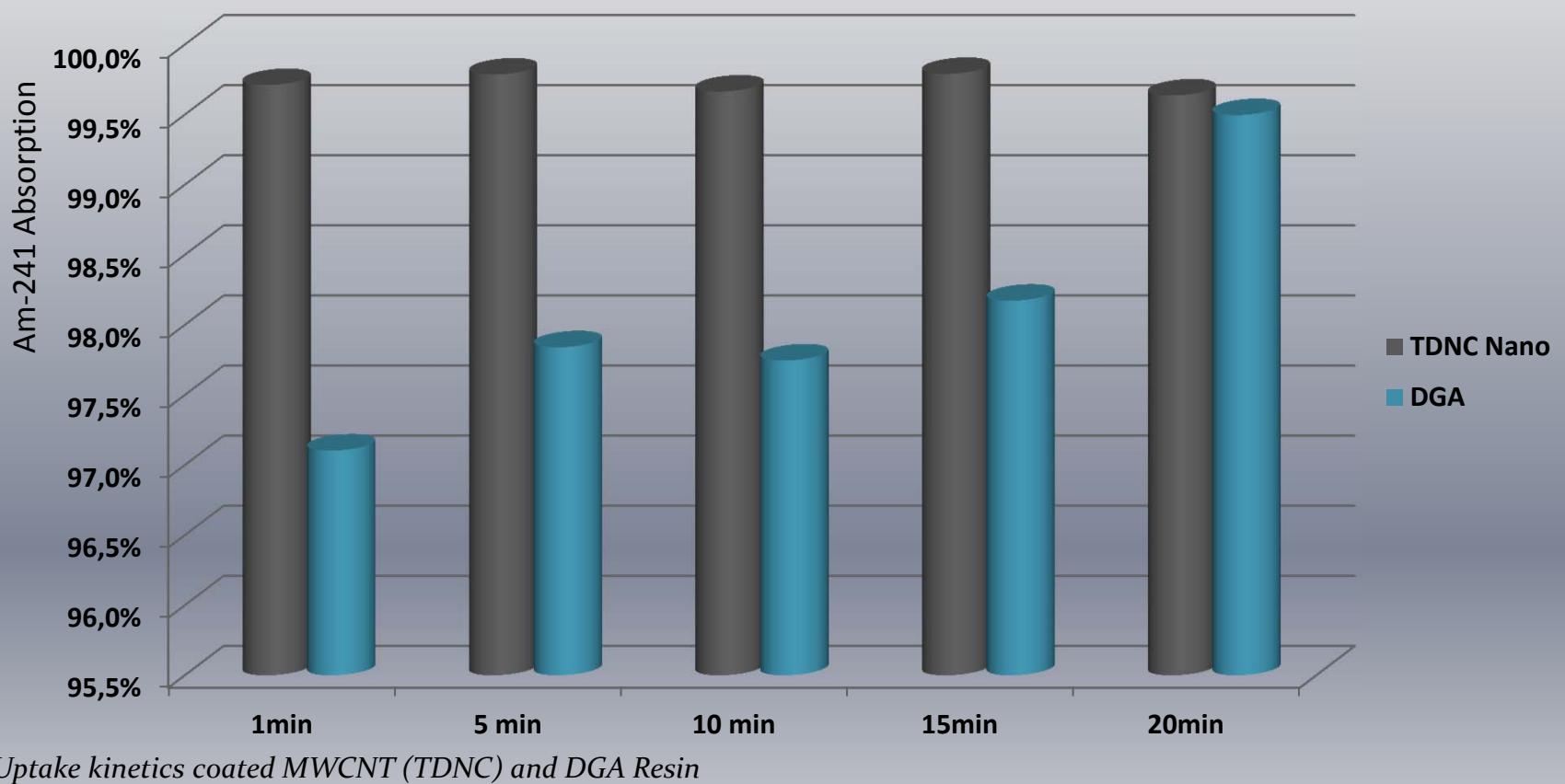
# Reproducibility of TDNC coating



Comparison Am  $D_W$  for different coated MWCNT (TDNC) lots

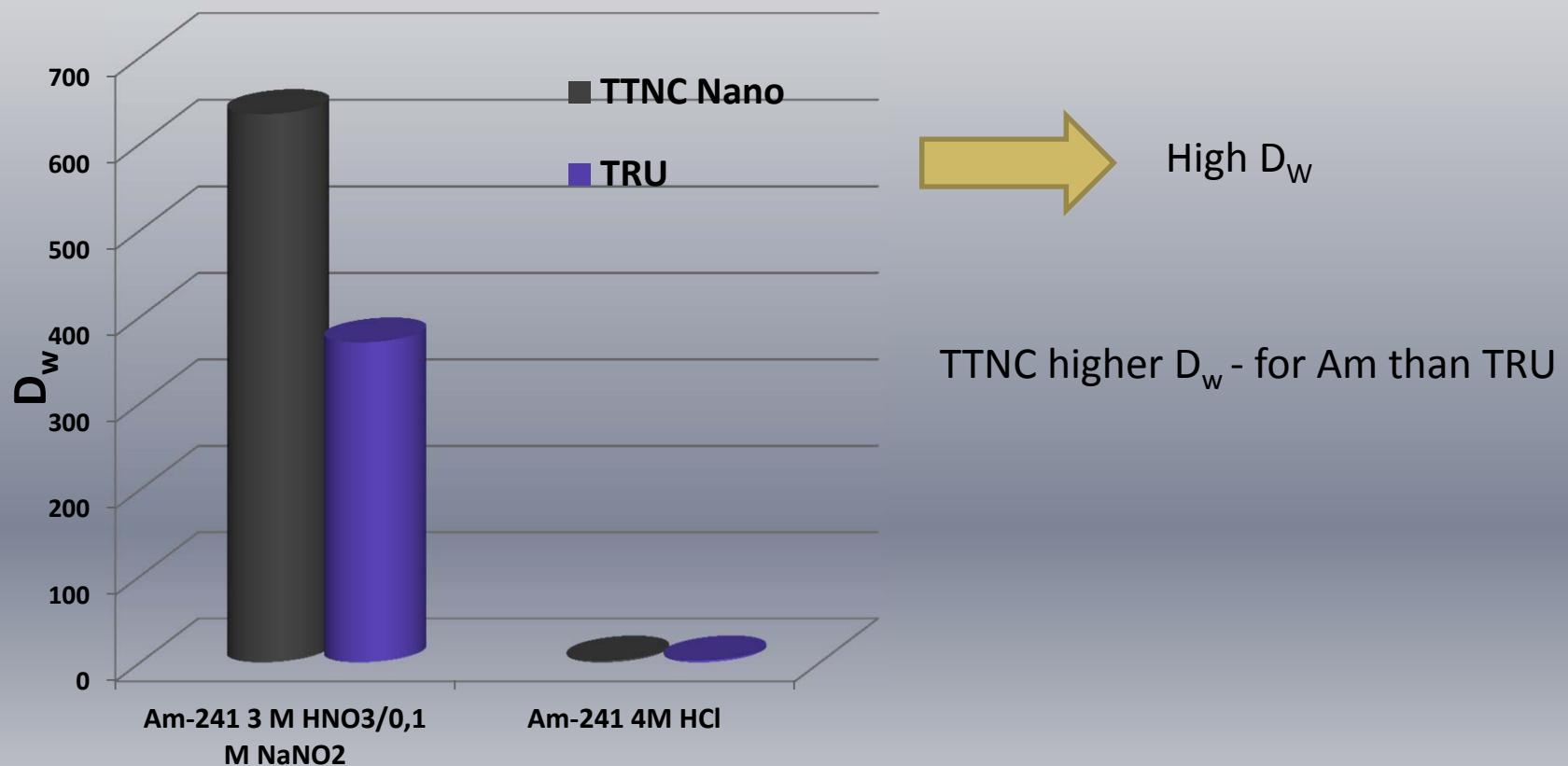
- Coating reproducible

# Kinetics: TDNC vs. DGA



**Very fast kinetics: eq. after < 1min**

# TTNC vs. TRU



Comparison Am  $D_w$  coated MWCNT (TTNC) and TRU Resin

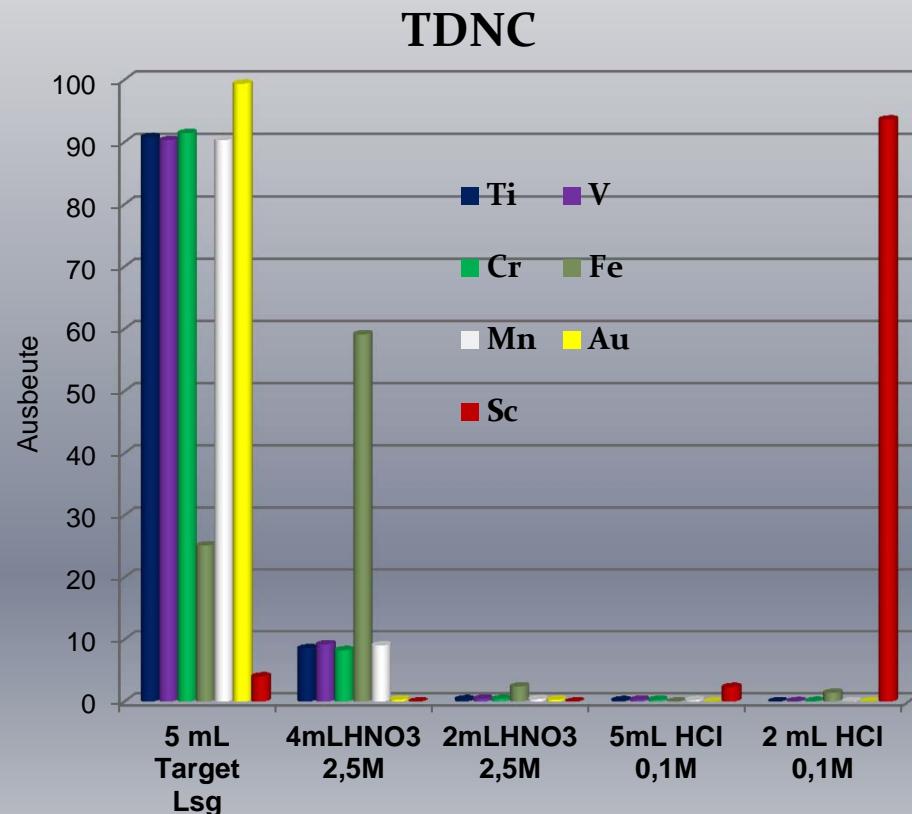
# Problems of working with Nanotubes

- Transfer from batch experiments to column geometry very difficult
- Low flow-rates
- Finding suitable filter / pore sizes
- Some solutions:
  - Special packing technique
  - Adequate poresize identified
  - Flowrates between 1-5 mL possible (using vacuum)

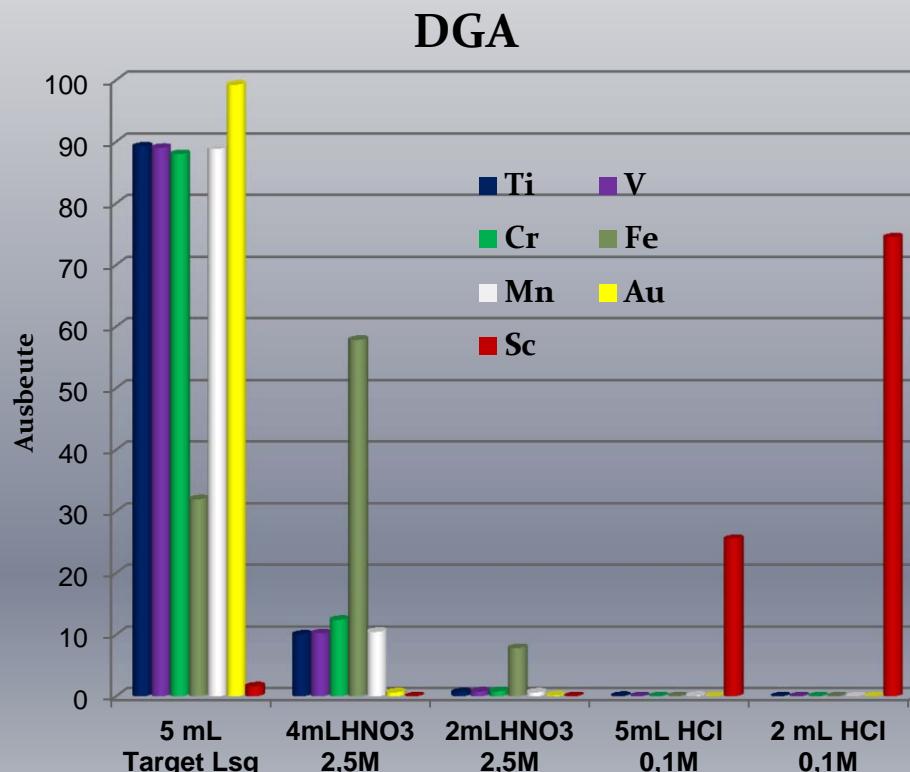
# Conclusions I

- ✓ TDNC and TTNC show very promising  $D_w$  in batch experiments
- ✓ Very rapid kinetics
- ✓ Column experiments possible
- ✓ Suitable flow-rates achievable

# TDNC vs. DGA simulated Ti target



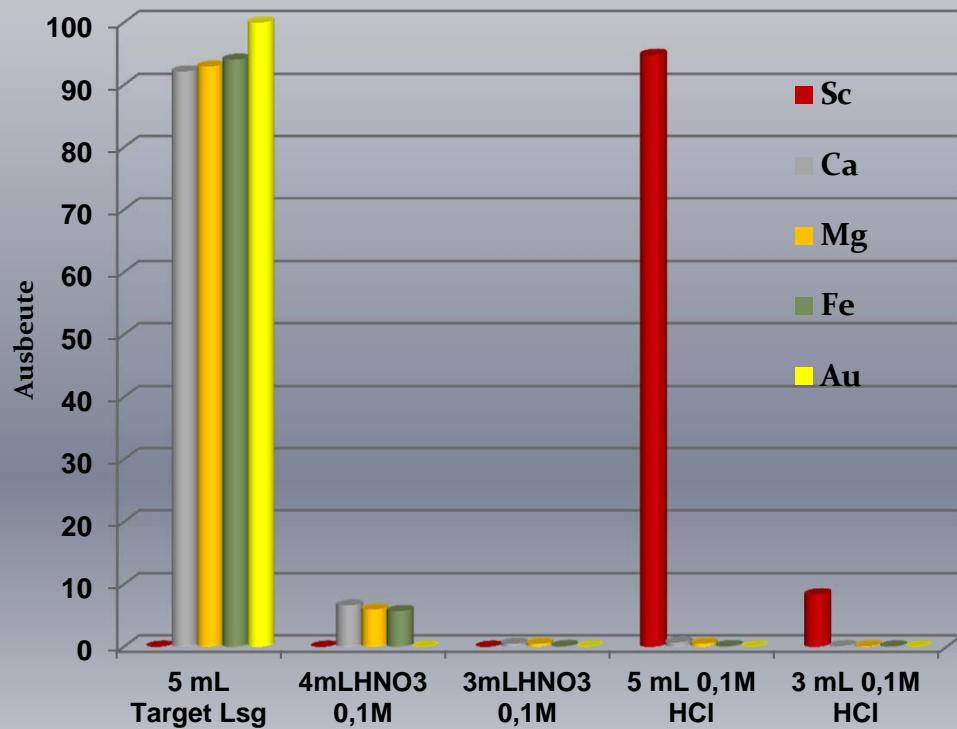
Elution study, simulated Ti target, coated MWCNT (TDNC)



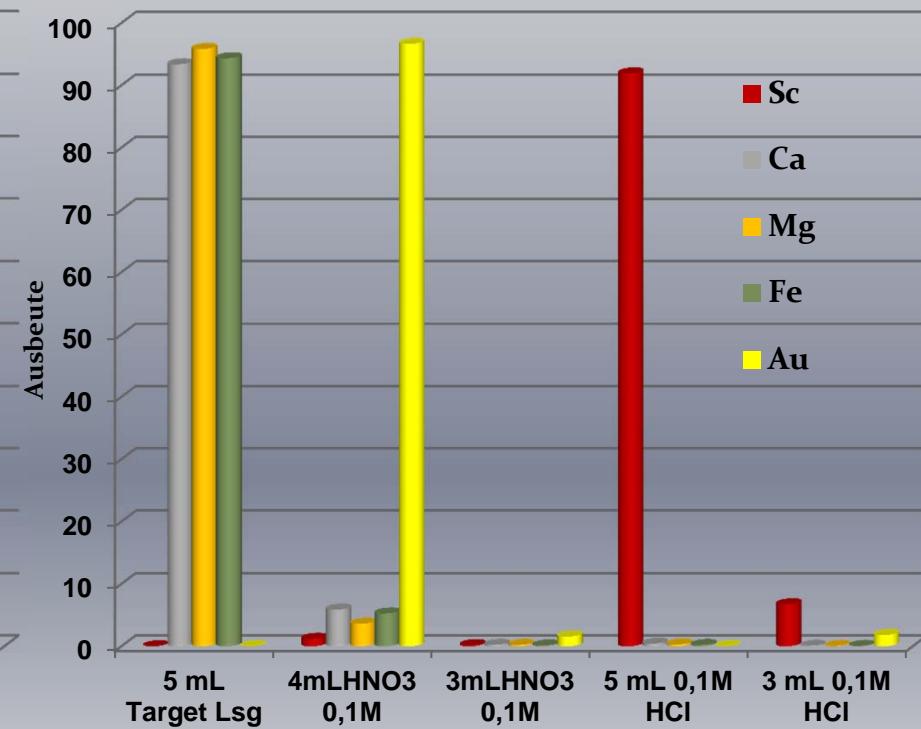
Elution study, simulated Ti target, DGA resin

# TDNC vs. DGA simulated Ca target

TDNC



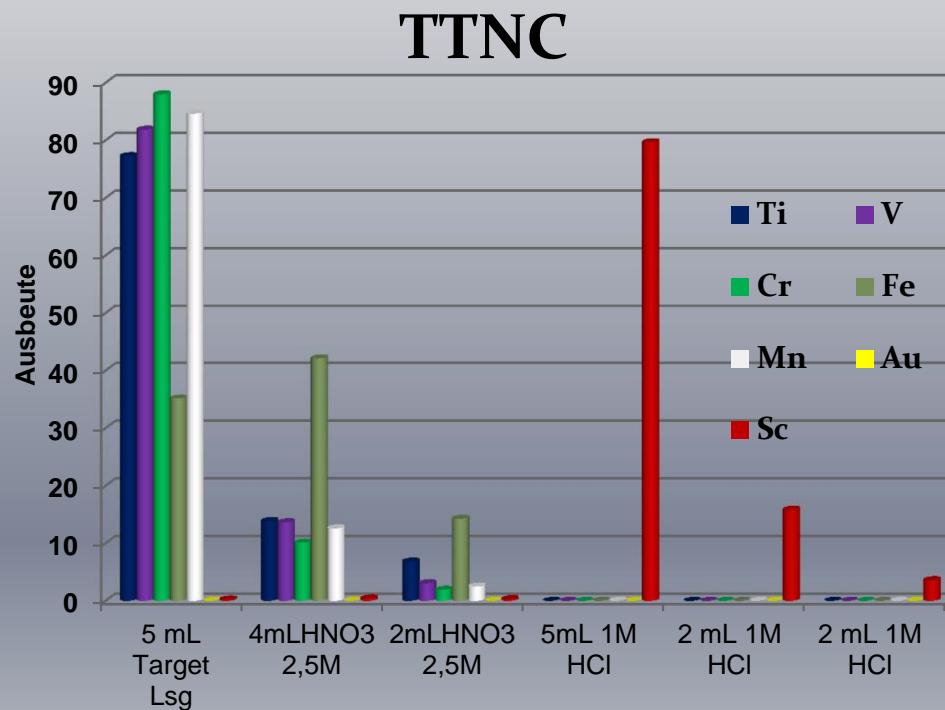
DGA



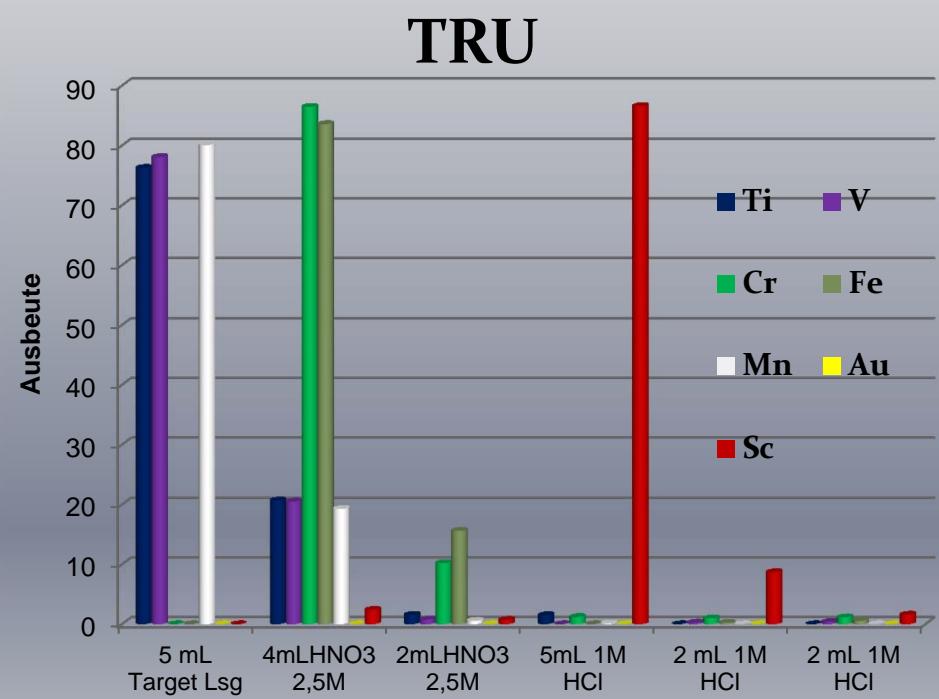
Elution study, simulated Ca target, coated MWCNT (TDNC)

Elution study, simulated Ca target, DGA resin

# TTNC vs. TRU simulated Ti target

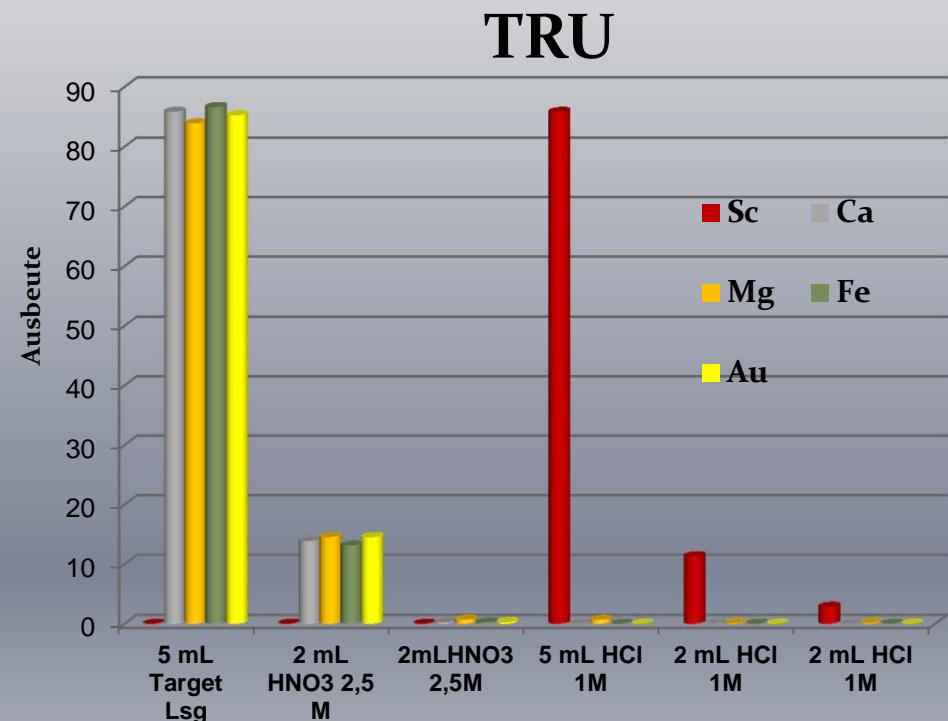
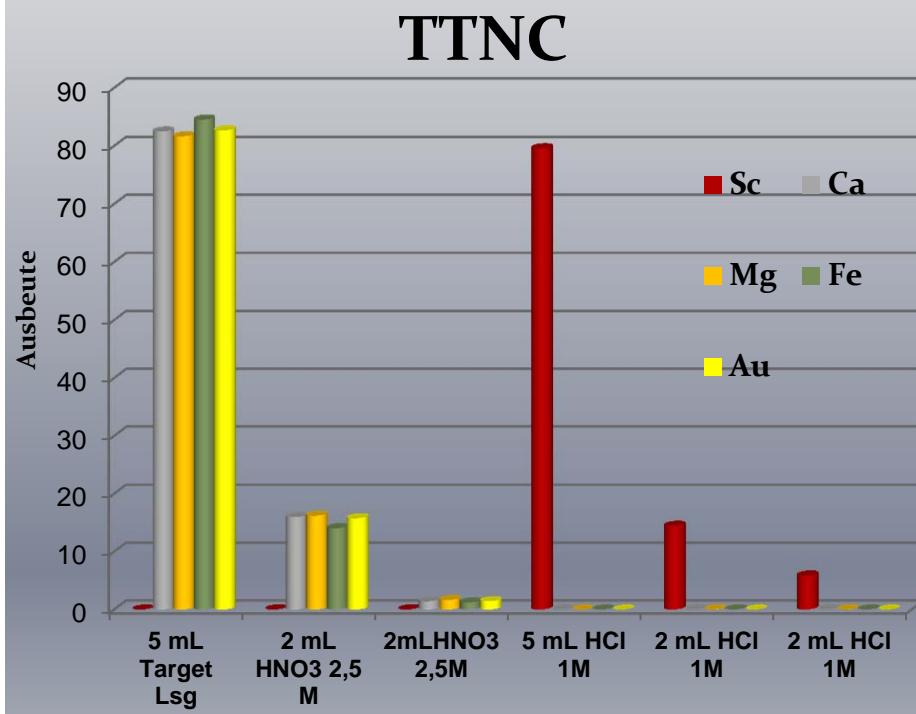


Elution study, simulated Ti target, coated MWCNT (TTNC)



Elution study, simulated Ti target, TRU resin

# TTNC vs. TRU simulated Ca target



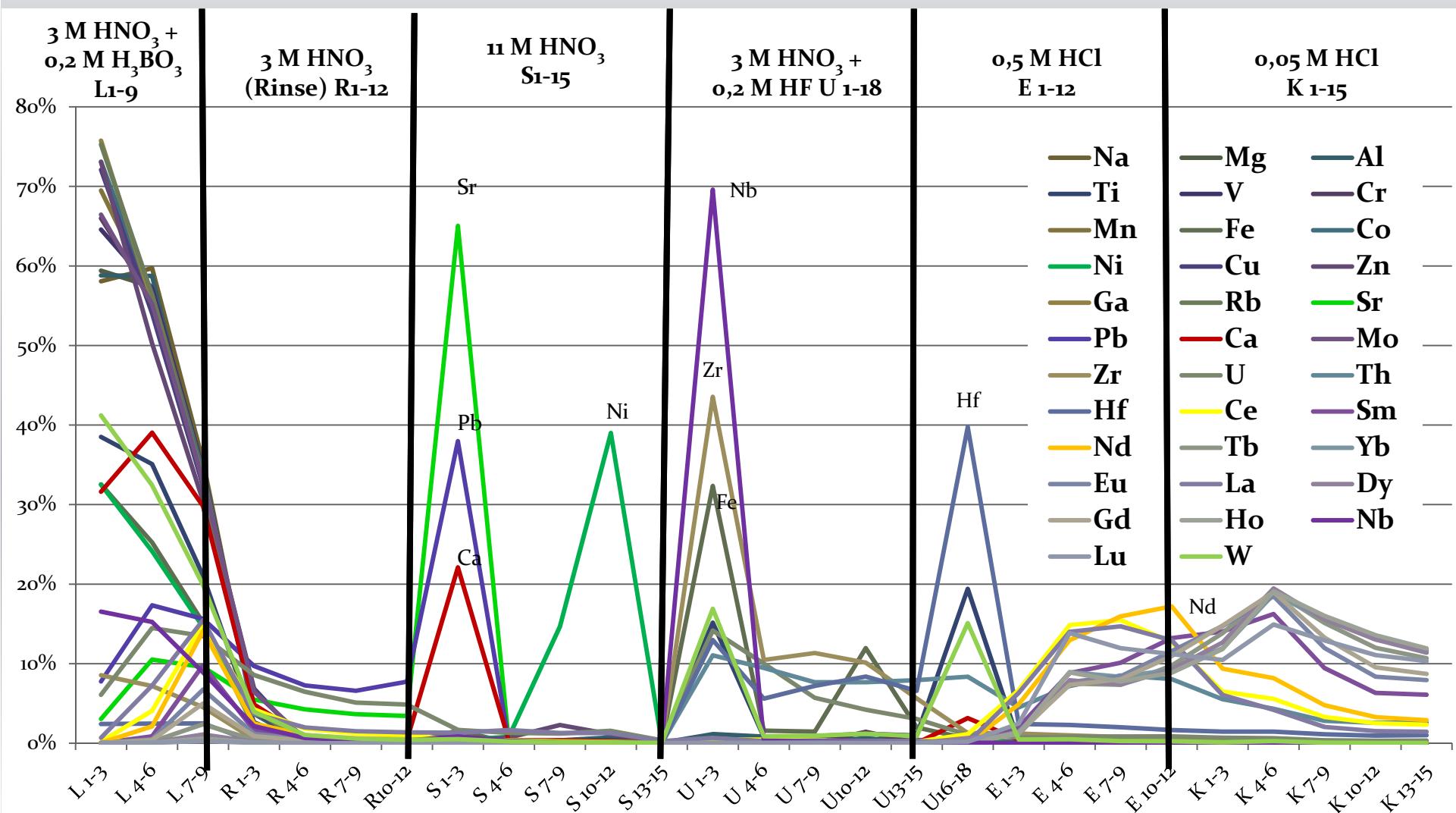
Elution study, simulated Ca target, coated MWCNT (TTNC)

Elution study, simulated Ca target, TRU resin

# Conclusions II

- Developed separation methods for Scandium (DGA and TRU) directly applicable to coated MWCNT
- Very good separation performance in column tests
- Adequate flow rates

# Complex separation on TDNC



# Perspectives

➤ Optimisation:

- Flow rate
- Extraktant / Nanotube ratio
- Column volume

➤ Improvement / variation of column material

# Thank you very much

- ❖ Prof. Jungclas, Kernchemie Marburg
- ❖ TrisKem International

# Optimized method

➤ Vacuum supported flow (1 – 3 mL/min)

➤ TDNC resin (400 mg)

- Load from 5 mL 2,5 M HNO<sub>3</sub> (Ti) or 0,1M HNO<sub>3</sub> (Ca)
- Rinse with 4 mL and 3 mL 2,5 M HNO<sub>3</sub>
- Load and rinse contain ~100% Ti bzw. Ca
- Sc elution in 5 and 3 mL 0,1M HCl

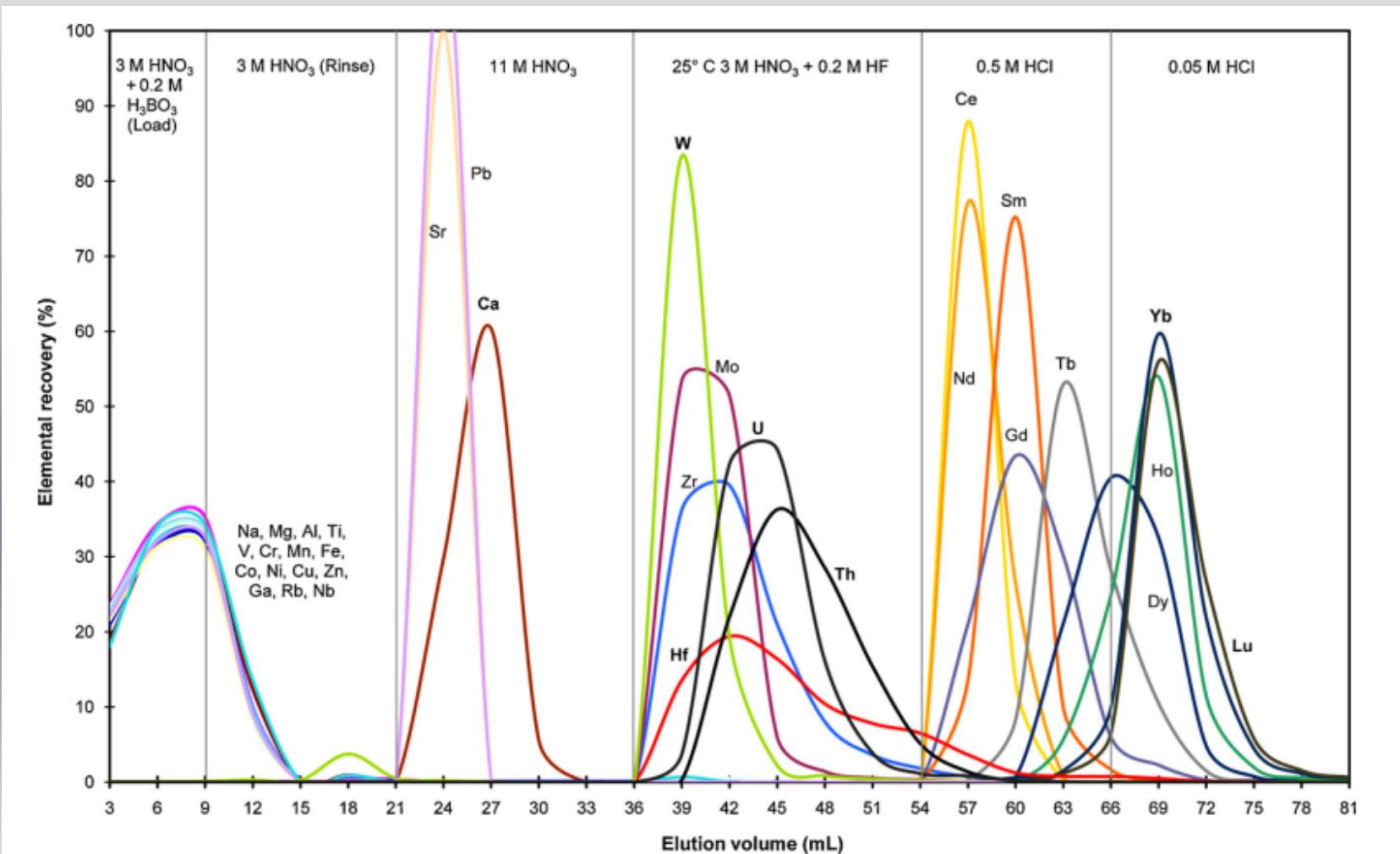
➤ TTNC resin (400 mg)

- Load from 5 mL 2,5 M HNO<sub>3</sub>
- Rinse with 2 x 2 mL 2,5 M HNO<sub>3</sub>
- Load and rinse contain ~100% Ti bzw. Ca
- Sc elution in 5 and 2 mL 1 M HCl

➤ Sc Yield > 98%

➤ Separation time: 10 min

# TODGA



doi:10.1016/j.talanta.2010.01.008